ARM Summer Training and Science Applications: Aerosol Radiative Forcing in Clear and Cloudy Skies





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GEOS-Chem Aerosol Optical Depth by Type











Rationale for Studying Aerosol

◇ Important in biogeochemical cycles (e.g. Sulfer cycle)
◇ Direct Effects on Visibility
◇ Effects on Human Health
◇ Water Cycle: Effects on Clouds and Precipitation Formation
◇ Radiative Forcing of Climate

IPCC AR5 Radiative Forcing





Aerosol Radiative Forcing according to IPCC AR5

Aerosol Direct Effect – aerosol radiative forcing in clear skies Aerosol Indirect Effect – aerosol radiative forcing in cloudy skies



What role does aerosol play in the climate system?

Aerosol Direct Effect – aerosol radiative forcing in clear skies Aerosol Indirect Effect – aerosol radiative forcing in cloudy skies



What about 'transition zones'? the integrated aerosol cloud system? ERF ARI+ACI

Top of the Atmosphere RF



Atmospheric RF

Surface RF

Global – Regional – Local Annual – Diurnal - Instantaneous

TOA Global Annual Average does not tell the whole aerosol story!

The cloudy-clear transition zone (a.k.a.'twilight zone', 'halos')



Varnai and Marshak 2009

Aerosol Direct Radiative Forcing



What are uncertainties for estimating aerosol radiative forcing?

The AeroCom Project: Aerosol comparisons between observations and models





MISR Team, Jet Propulsion Lab/Caltech and NASA Goddard Space Flight Center

Aerosol optical depth horizontal and vertical distribution 8-year average: IPCC AR5



What are uncertainties for estimating aerosol radiative forcing?

spatial distribution (horizontal and vertical) of aerosol amount and properties

• aerosol formation processes: emissions to burden

- composition & size distribution to optical properties
- optical properties to radiative fluxes

feedbacks 🖌

 aerosol heating and cooling effects on circulation, cloud development, temperature

Pathways to Calculating Aerosol Direct Radiative Forcing



 $\alpha_{abs} = \log(\sigma_{abs 450 \text{ nm}} / \sigma_{abs 700 \text{ nm}}) / \log(450/700)$

extinction efficiency: $Q_e = total mass/(\sigma_{sca} + \sigma_{abs}) m^2 g^{-1}$

Pathways to Aerosol Direct Radiative Forcing: Observed and Modeled



n+*i*k

AeroCom Diversity in optical properties



What is an aerosol? A liquid or solid particle suspended in a gas Originate from natural and anthropogenic sources



courtesy Jessie Creamean



Primary Formation Directly emitted

courtesy Jessie Creamean

Aerosol Mixing State

External Mixture Internal Mixture Soot Organic matter real world Aging Processes condensation Soot Adachi et al. 2010 coagulation Sulfates chemical transformation model world Aging Processes

'parameterization'

the Mixing State Index χ





E - Marine particles collected at Pt. Reyes National Shore Park, California. Red and yellow colors indicate the internal distribution of two different oxidation states of sulfur (S(IV) and S(VI)) inside individual particles (Liu et al. 2011b) F - Atmospheric dust particles collected on Okinawa Island, Japan. Color scale indicates fraction of Fe(II) present in individual particles (Moffet et al. 2012) G - Marine particles (blue) internally mixed with anthropogenic organic material (green) collected in vicinity of Sacramento, California (Laskin et al. 2012b)

Aerosol Morphology





[China et al., 2013]



Soot morphology and mixing: Sacramento 2010

courtesy Claudio Mazzoleni



volcanic ash

pollen

sea salt

soot



courtesy N. Riemer



Aerosol size distributions: function of emissions, formation, and processing



Size is the most fundamental parameter for determining aerosol optical properties and radiative effects.

Aerosol Nucleation New Particle Formations events at SGP



courtesy Peter McMurry and Jim Smith

Aging Processes



- some primary particles are initially insoluble
- can become partly soluble through
 - coagulation
 - chemical transformation
 - condensation
- results in
 - varied optical properties
 - greater CCN activity

Growth and coagulation



Aging Processes



Rate of ageing: time to form 1 monolayer (~10¹⁹ molec m⁻²) of H_2SO_4 on a 100 nm particle

- ~10⁴s (3 h) in <u>polluted conditions</u> ($H_2SO_4 = 10^7 \text{ cm}^{-3}$)
- ~10⁵ 10⁶ s (1-10 days) in <u>clean conditions</u>
- Factor >10 longer for coarse particles

these processes coupled with meteorology determine the spatial distribution of aerosol and their radiative properties

Hygroscopicity measure of water uptake by soluble particles

- Reversible process (deliquescence, effloresence)
- f(RH, composition)
- Timescale for water equilibration ~ seconds



Petters and Kreidenweis, 2007

Particle activation in clouds



Size is the most fundamental parameter for aerosol radiative effects.



Bioaerosol sources

- "Bioaerosol" comprises particles from viruses (10's of nm) to pollen and leaf fragments (100's of um)
- Some are emitted actively and some passively



Pollen > ~10 um

Bacteria

~1 um

Fungal spores: 2-10 um





• Most sampling at 300m, some excursions up to 1 km.

Clou

 In addition to the WIBS, the blime as equipped with:







Perring et al, JGR, 2015

Global distribution of aerosol chemical composition



Zhang et al. 2007



Biomass Smoke over South America



Vertical Distribution of Aerosol: Impacts on Radiative Forcing



Aerosol-radiation interactions

Scattering aerosols



(a)



Aerosols scatter solar radiation. Less solar radiation reaches the surface, which leads to a localised cooling.



The atmospheric circulation and mixing processes spread the cooling regionally and in the vertical.





Aerosols absorb solar radiation. This heats the aerosol layer but the surface, which receives less solar radiation, can cool locally.



At the larger scale there is a net warming of the surface and atmosphere because the atmospheric circulation and mixing processes redistribute the thermal energy.

Absorbing aerosols

Optical and Radiative Properties: Standard Long-Term Products

Measured quantities	Derived quantities
$\sigma_{\rm sca}$ = scattering coefficient Mm ⁻¹	extinction efficiency:
$\sigma_{ m abs}$ absorption coefficient Mm ⁻¹	$Q_e = total mass/(\sigma_{sca} + \sigma_{abs}) m^2 g^{-1}$
$\sigma_{\rm bsca}$ = backscattering coefficient Mm ⁻¹	single scattering albedo: $\omega_0 = \sigma_{sca}/(\sigma_{sca} + \sigma_{abs})$
aerosol mass concentration (total/ speciated) μ g m ⁻³	asymmetry parameter:
surface radiative fluxes W m ⁻²	$g = 1.011 - 1.036\beta + 2.0059\beta^2,$ $\beta = 0.0817 + 1.8495\sigma_{bsca} - 2.9682\sigma_{bsca}^2$
Radiative Forcing	scattering Ångström exponent:
$RF = (f_{\alpha} \mathbf{\Psi} - f_{\alpha} \mathbf{\uparrow}) - (f_{0} \mathbf{\Psi} - f_{0} \mathbf{\uparrow}) W \; m^{-2}$	$\alpha_{\rm sca} = \log(\sigma_{\rm sca 450 nm} / \sigma_{\rm sca 700 nm}) / \log(450 / 700)$
$(f \Psi - f \uparrow) = \text{net irradiance (upwelling minus)}$	absorption Ångström exponent:
downwelling flux)	$\alpha_{abs} = \log(\sigma_{abs 450 nm}/\sigma_{abs 700 nm})/\log(450//00)$
$f_{a} = $ flux with aerosol	humidification factor:
$t_0 = $ flux without aerosol	fRH = $\sigma_{sca 85\% RH} / \sigma_{sca 40\% RH}$
RFE = RF / $\tau_a W m^{-2}$	sub-micron scattering fraction:
$\tau_{a} \sim Q_{e}$	$R_{\rm sca} = \sigma_{\rm sca 1 \ \mu m} / \sigma_{\rm sca 10 \ \mu m}$

ARM Aerosol Measurements



number & size

number conc CPC 3772 (> 10 nm) CPC 3776 (> 2.5 nm)

size distribution 50 nm – 1 μm UHSAS

size distribution 400 nm – 15 μm APS

CCN number conc. spectrum with S CCN-2

growth factor HTDMA optical properties

dry light extinction CAPS 3λ

light scattering nephelometer 3λ

light absorption PSAP (CLAP) 3λ

humidified light scattering nephelometer + humidograph 3λ aerosol chemical composition

non-refractory mass concentration (ToF-)ACSM gas phase chemistry

CO Los Gatos

CH₄, CO₂ Picarro

SO₂ ThermoScientific

 NO_x/NO_y NO_x Analyzer

O₃ Ozone Analyzer