



# Cloud Condensation Nuclei Closure During CLASIC 2007

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 Office of  
Science  
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# Introduction

## Introduction

Methodology

CLASIC 2007

Summary

- A major redesign of the Texas A&M Differential Mobility Analyzer/Tandem Differential Mobility Analyzer (DMA/TDMA) occurred in 2005/2006
  - Simultaneous measurements of size distributions and the hygroscopic properties of the aerosol
- For the CLASIC campaign, a Droplet Measurement Technologies (DMT) Cloud Condensation Nuclei Counter (CCNc) was also operated on the aircraft to measure the concentration of CCN active at supersaturations of 0.165, 0.305, and 0.60 %

# Motivation

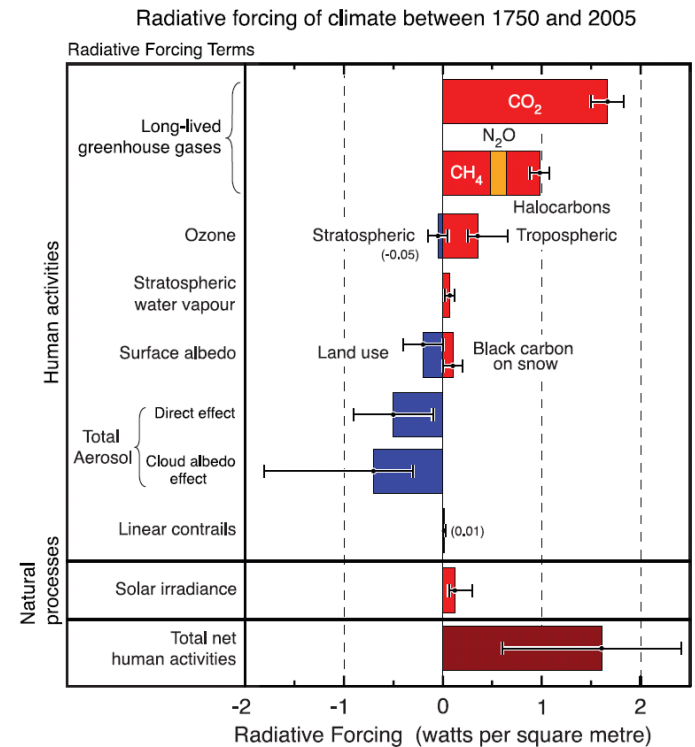
## Introduction

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- Characterize the aerosol size distributions and physicochemical properties in various environments
- Predict the effect on visibility and cloud condensation nuclei formation
  - Compare predicted versus measured CCN to ascertain our level of understanding particle activation
- Important to constrain results for global and regional climate models



IPCC 2007

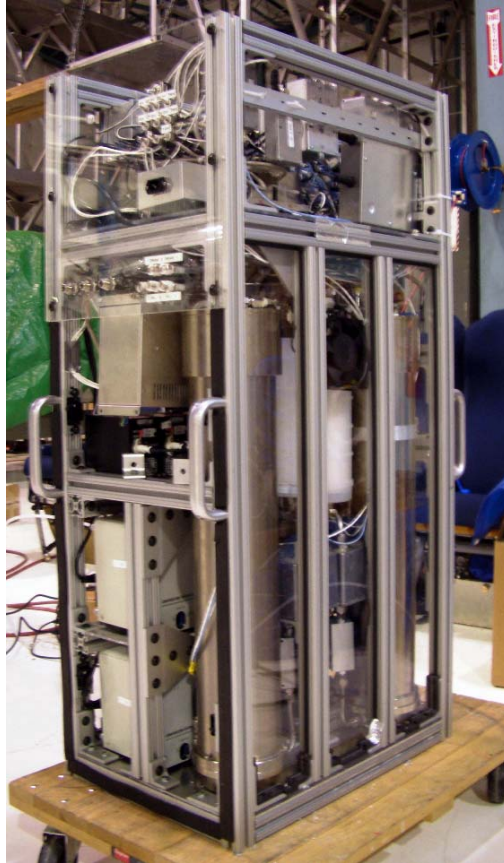
# DMA/TDMA

Introduction

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- 3 Hi-Flow Differential Mobility Analyzers
  - Operated at 10:1 flow ratio
- DMA size distribution measured from 0.010 to .700  $\mu\text{m}$  every 90s
- TDMA growth distributions for dry diameters of 0.025, 0.050, 0.100, 0.200, and 0.400  $\mu\text{m}$  measured every 12 minutes
- Maintained at 85% and 38°C throughout the project

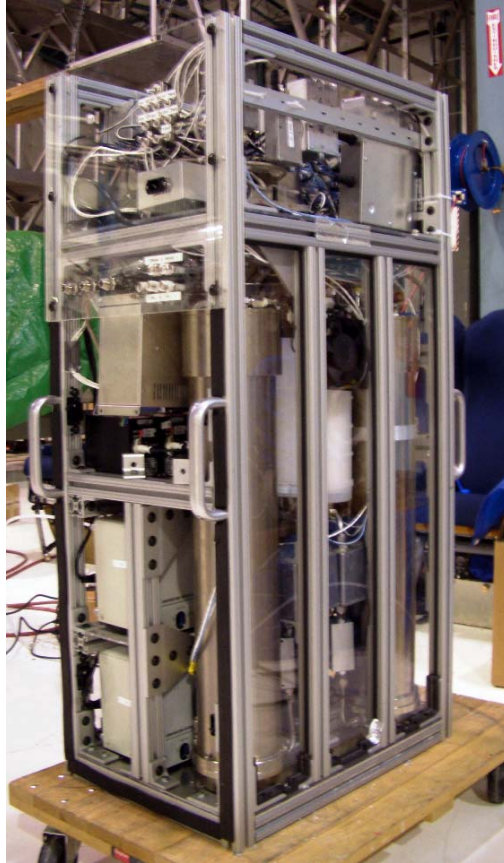
# DMA/TDMA

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**Methodology**

CLASIC 2007

Summary



- Calibrated at the beginning of CLASIC
- DMA
  - Inject PSL
- TDMA
  - Inject atomized ammonium sulfate and sodium chloride
  - Daily dry scans

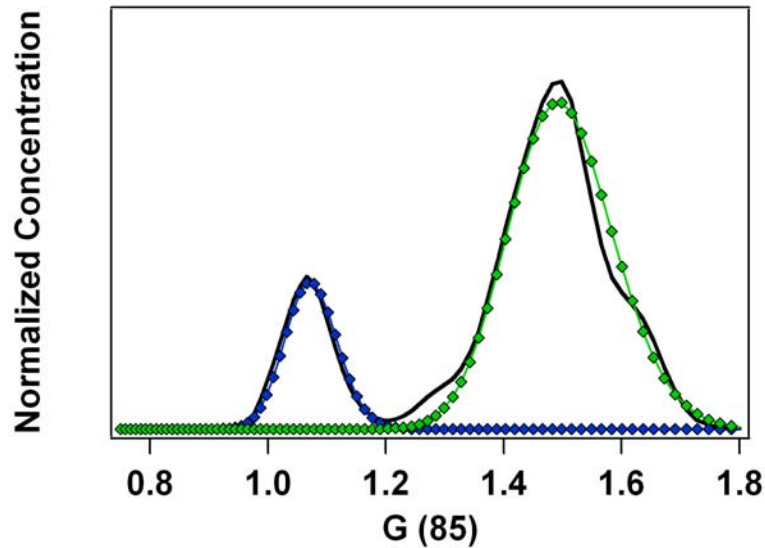
# DMA/TDMA

Introduction

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Summary



	Mode 1	Mode 2
<b>N</b>	0.24	0.76
<b>GF</b>	1.07	1.49
$\sigma$	1.04	1.06

- A log-normal distribution was fitted to each mode. The geometric means of the log-normals are assumed to represent the hygroscopicity of the aerosol population.

# DMT CCN<sub>c</sub>

Introduction

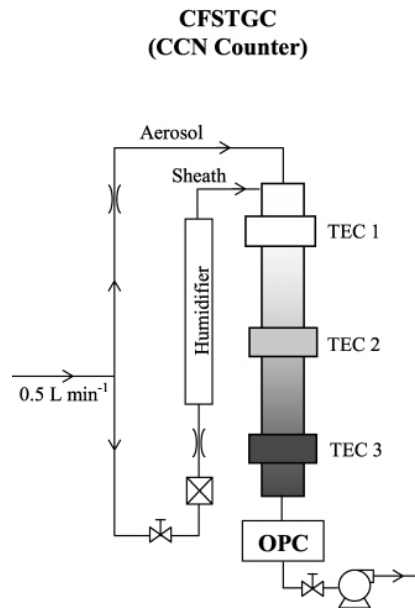
**Methodology**

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Summary

*Lance et al., 2006*

*Image Source: DMT*



- Atomized ammonium sulfate is used to calibrate



# CCN Prediction

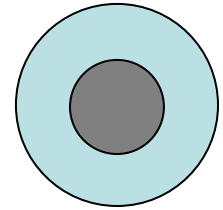
Introduction

**Methodology**

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- Use modified Köhler Theory to determine  $S_c$  value of a particle with known dry diameter and  $G(85)$
- Assumptions:
  - Particle is composed of an insoluble core surrounded by a soluble shell
  - Soluble material is composed of Ammonium Sulfate with a van't Hoff factor ( $i$ ) of 2.3
  - In subsaturated conditions, the vapor pressure over the particle is in equilibrium with ambient vapor pressure
  - Constant TDMA Temperature of 311 K





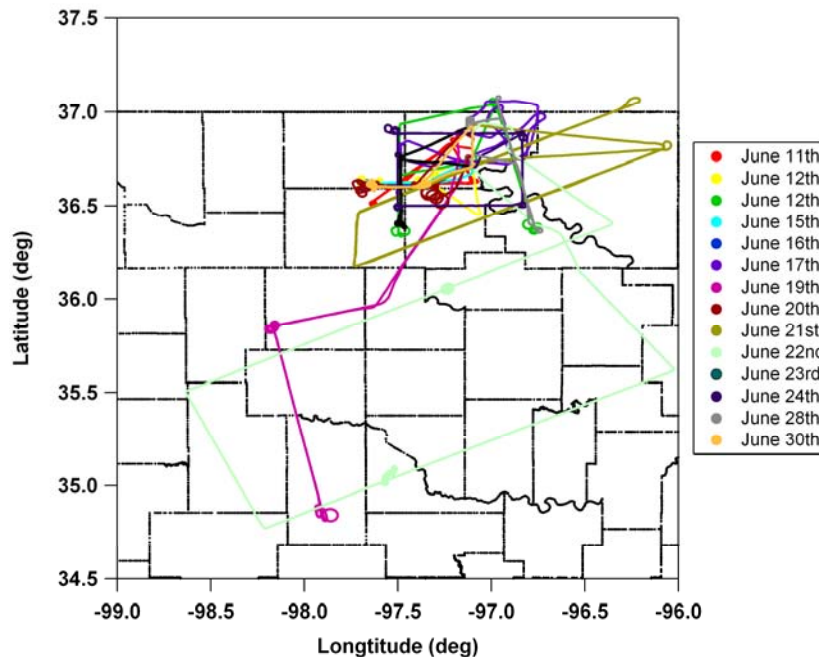
# CLASIC 2007

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Summary



- 14 research flights were flown to study how land surface processes affect atmospheric aerosol loading and chemistry and the resulting effects on the microphysical and macrophysical properties of cumulus cloud fields
- Measured over 1300 size distributions
- 110 hygroscopic growth factor distributions for each dry diameter of 0.025, 0.050, 0.100, 0.200, and 0.400  $\mu\text{m}$
- DMT CCNc operated simultaneously



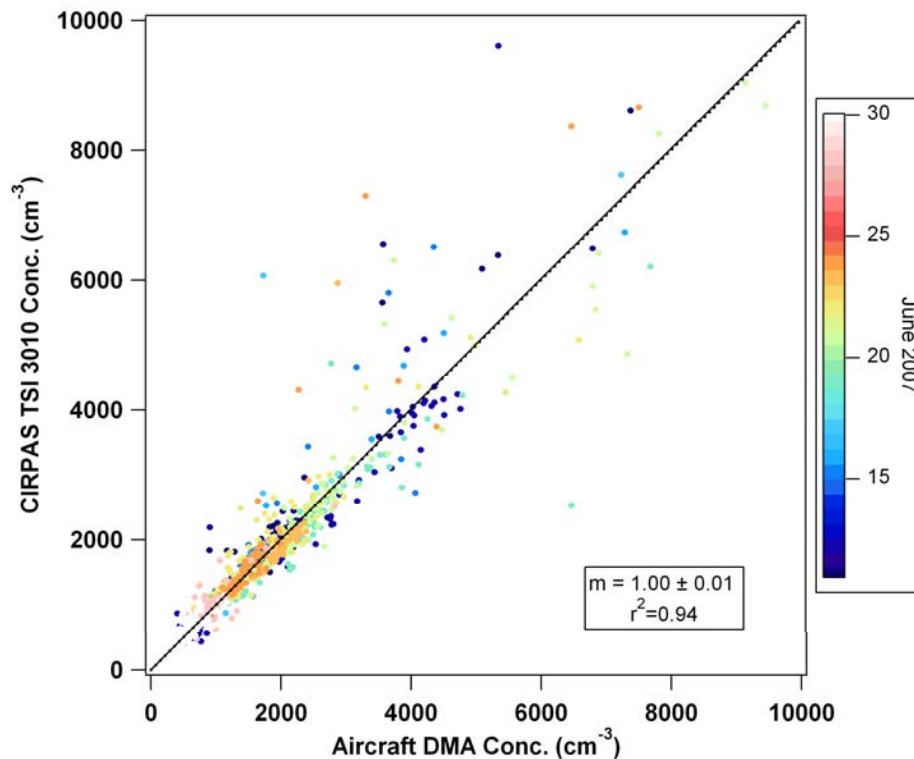
Source: PNNL

# CLASIC 2007

- Introduction
- Methodology
- CLASIC 2007**
- Summary

	June 2007													
	11	12	12	15	16	17	19	20	21	22	23	24	28	30
<b>Size Distributions</b>														
<b>G (85)</b>														
<b>0.165%</b>														
<b>0.305%</b>														
<b>0.600%</b>														

# Intercomparison



- Intercomparison between integrated number concentration and measured concentration from the CIRPAS TSI 3010

# DMT CCN<sub>c</sub>

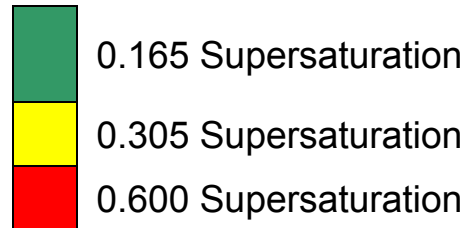
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- For most flights the CCN counter was operated as follows:

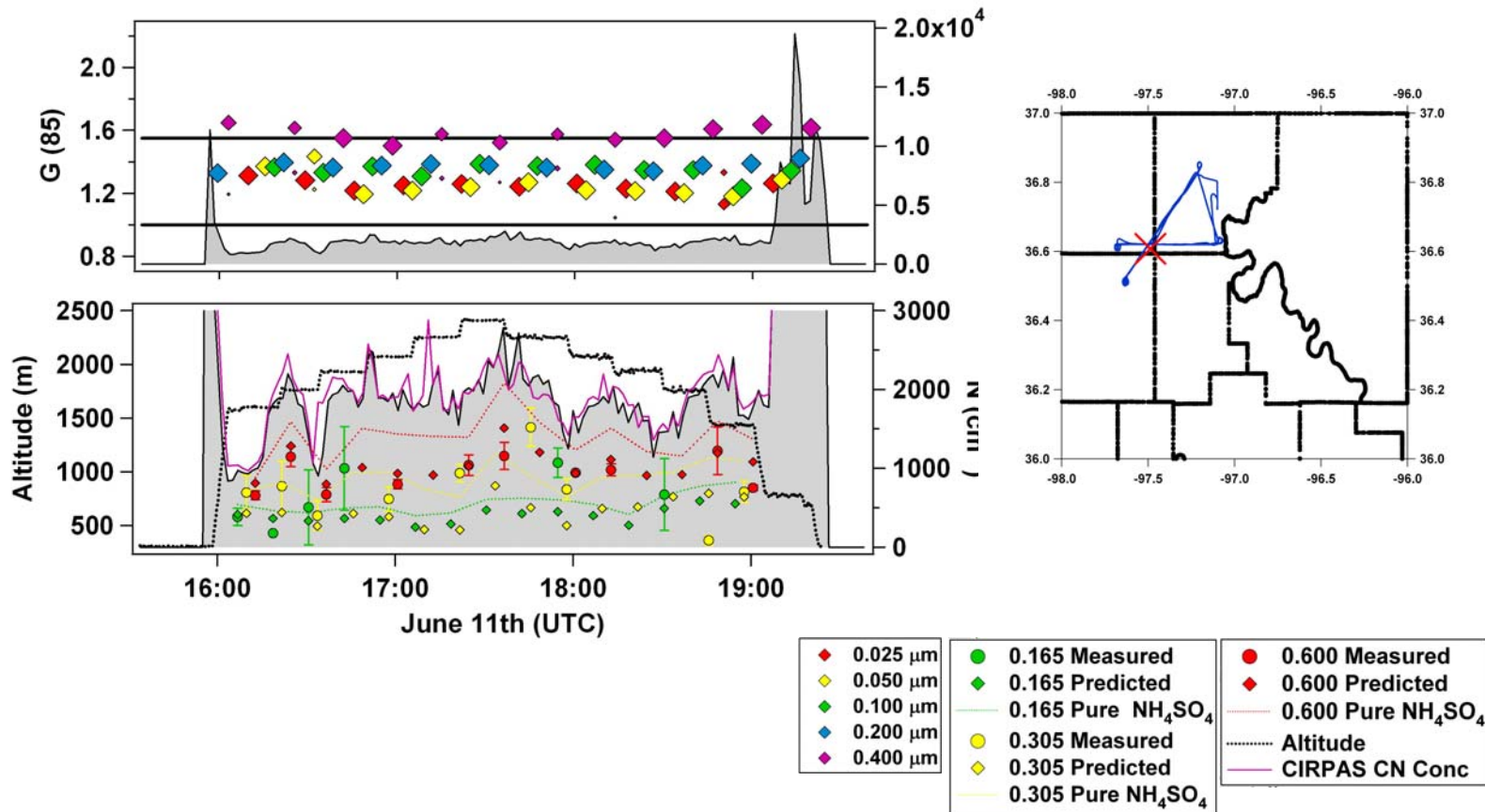


- Research flights with rapid altitude changes the supersaturation was fixed at 0.305 %
- Measured CCN concentrations were passed through several quality control checks
  - Are the flows within 10% of flows during calibration?
  - Does the first bin contribute a significant fraction to the overall counts?
  - Are the measured values greater than one would expect for pure ammonium sulfate?
- Number of data points after QC
 

– 0.165 %: n=22	0.305 %: n=37
Fixed 0.305%: n=33	0.600 %: n=49

# June 11<sup>th</sup> Time Series

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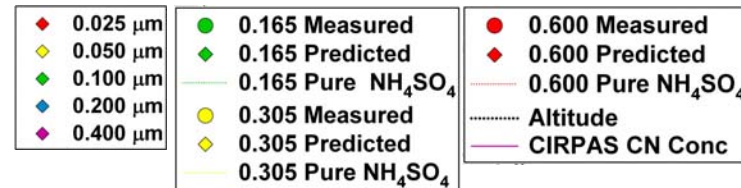
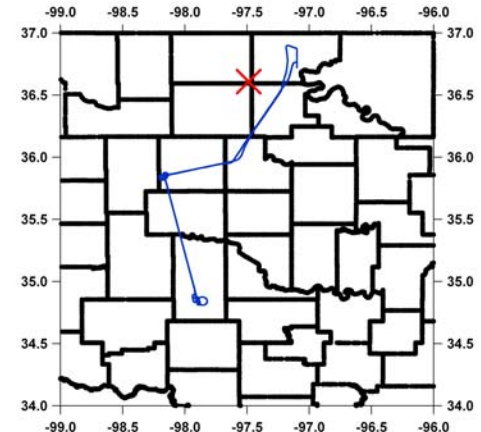
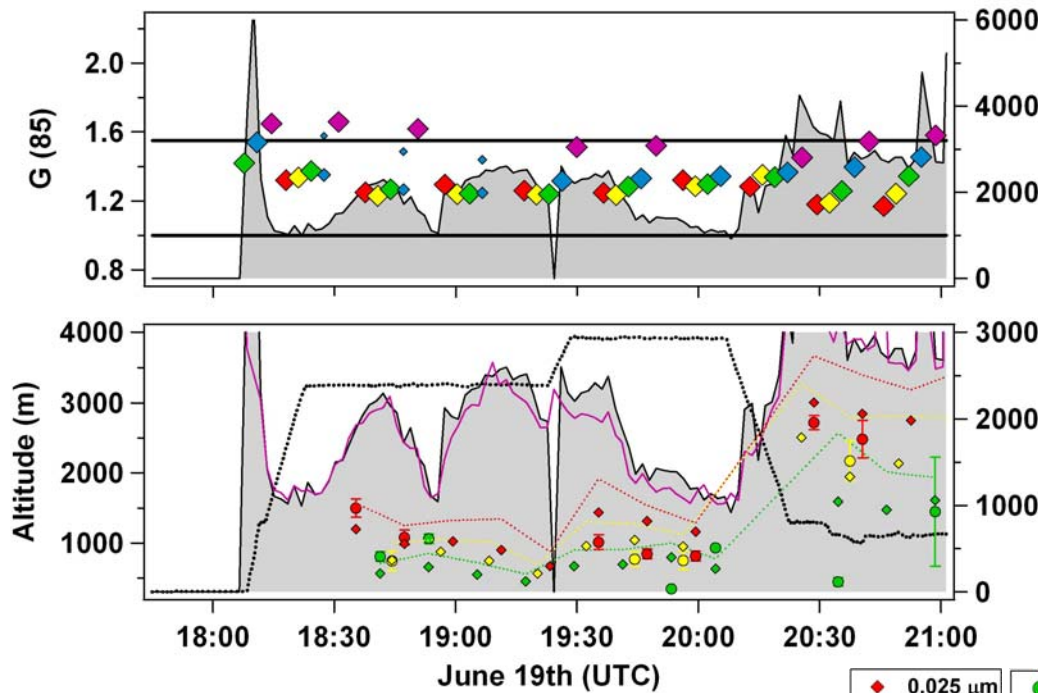


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# June 19<sup>th</sup> Time Series

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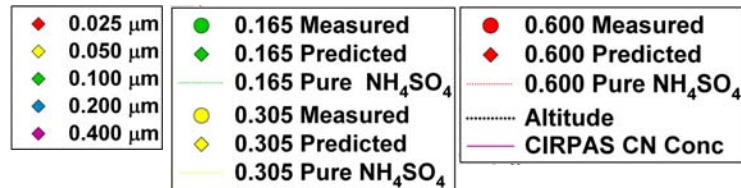
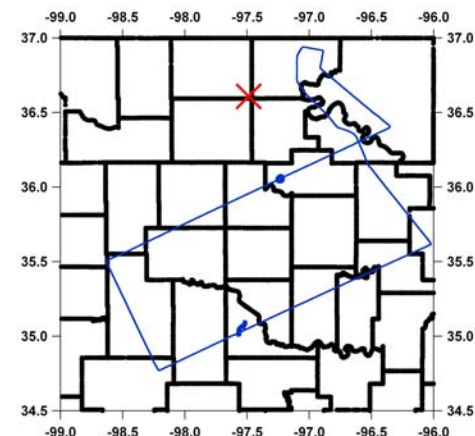
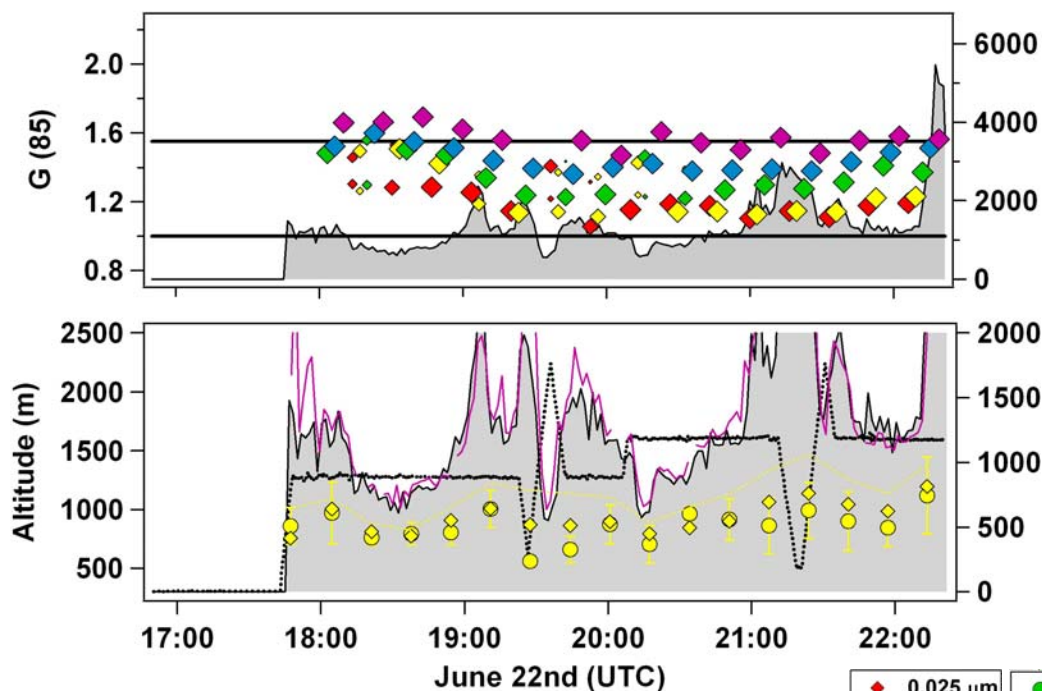


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# June 22<sup>nd</sup> Time Series

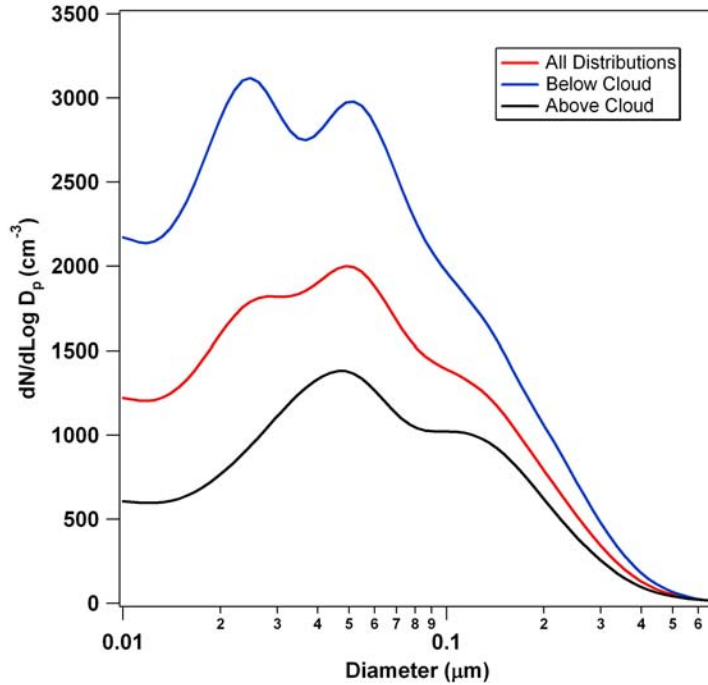
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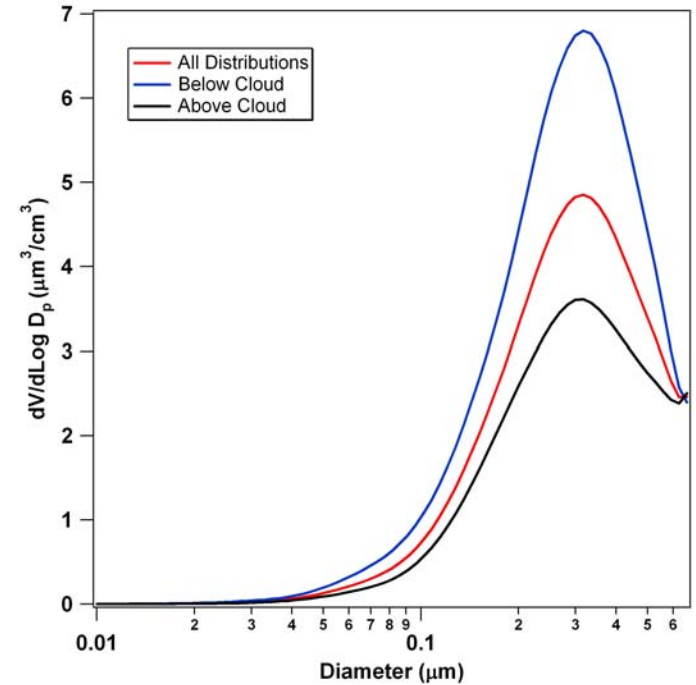
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# Size Distributions

### Number Distribution

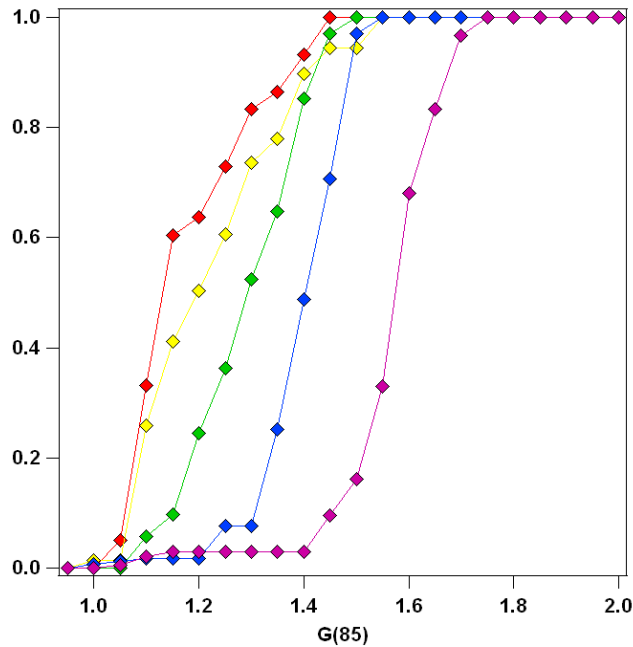


### Volume Distribution

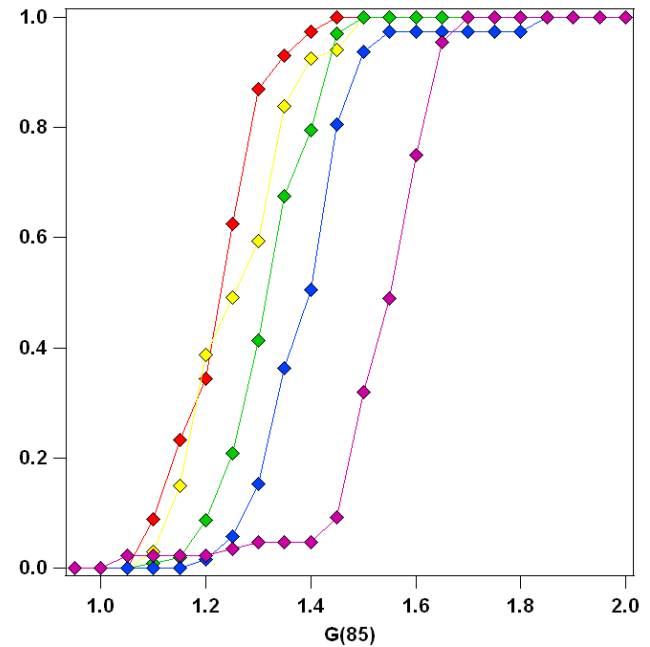


# G (85)

### Boundary Layer

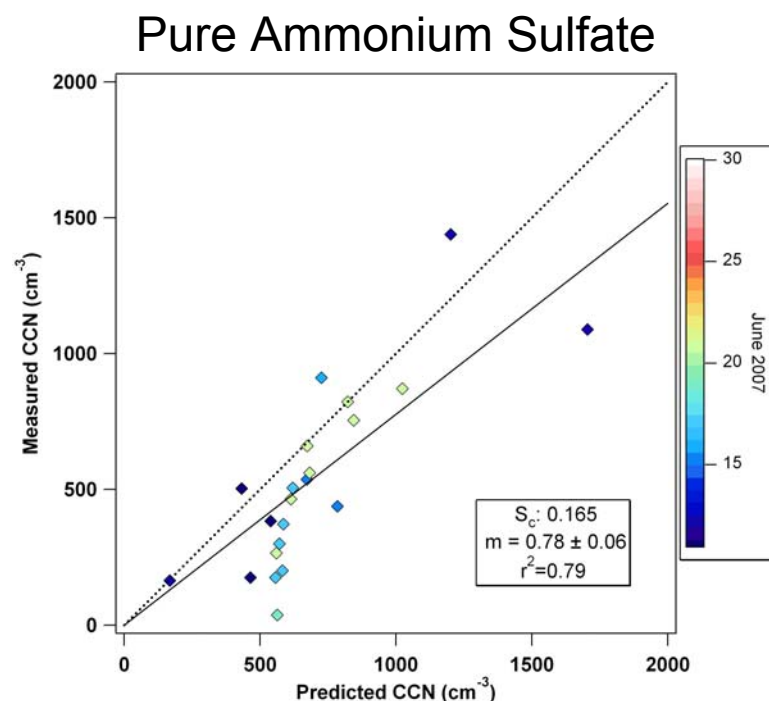
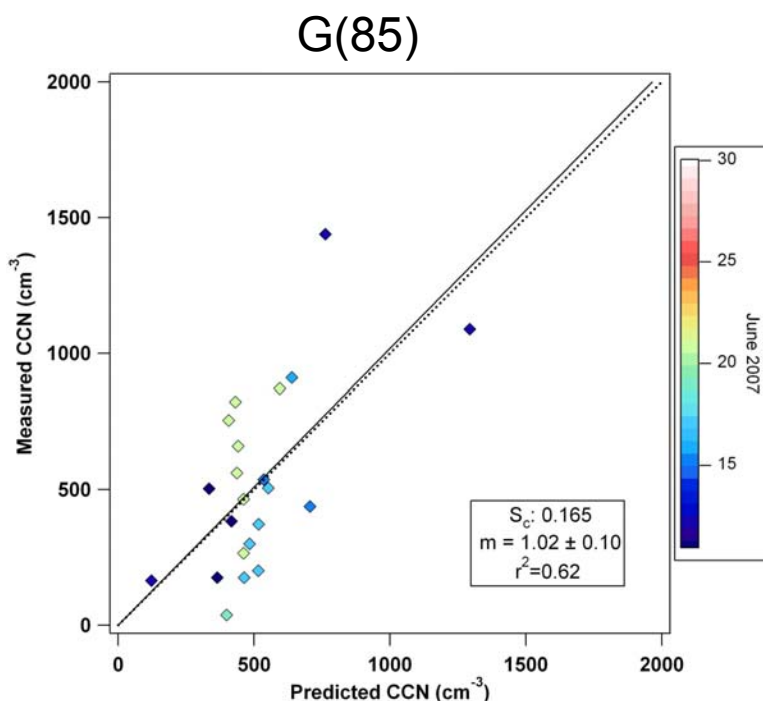


### Free Troposphere



# CCN Scatterplot (0.165)

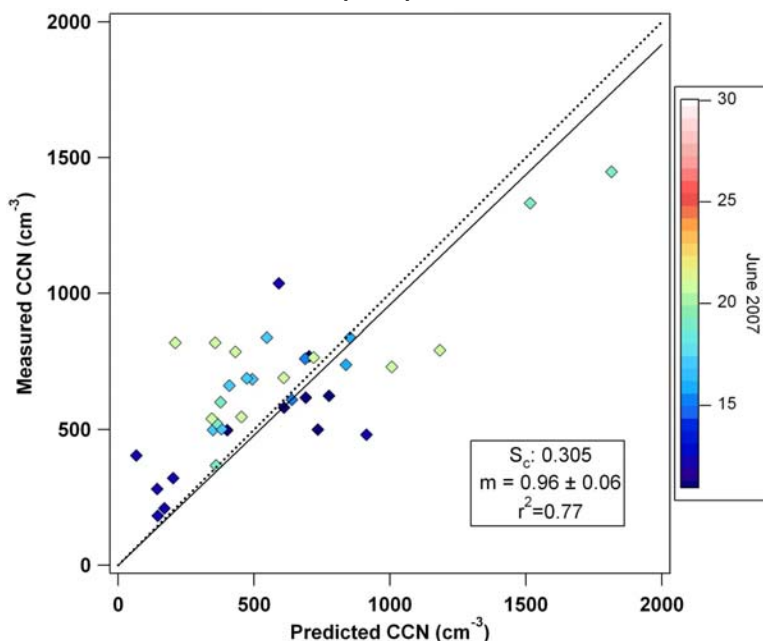
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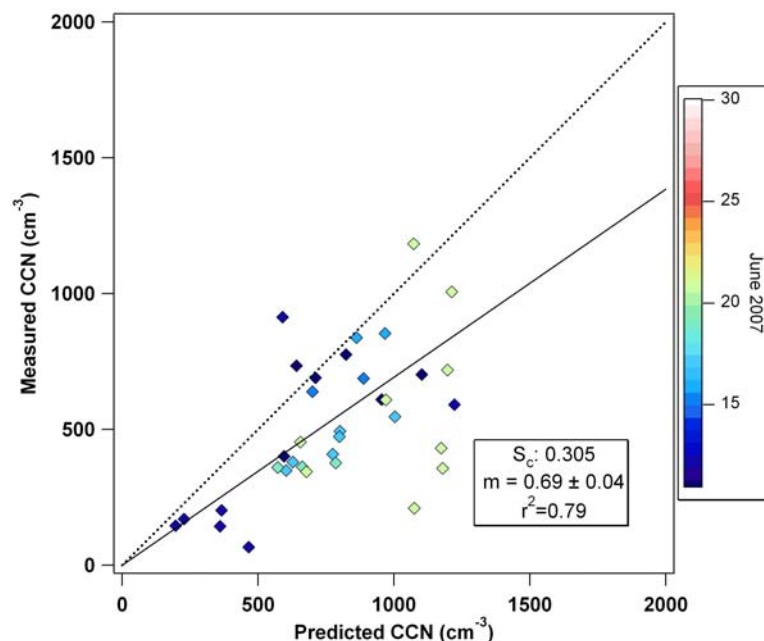
# CCN Scatterplot (0.305)

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G(85)



Pure Ammonium Sulfate



# CCN Scatterplot (0.305 Fixed)

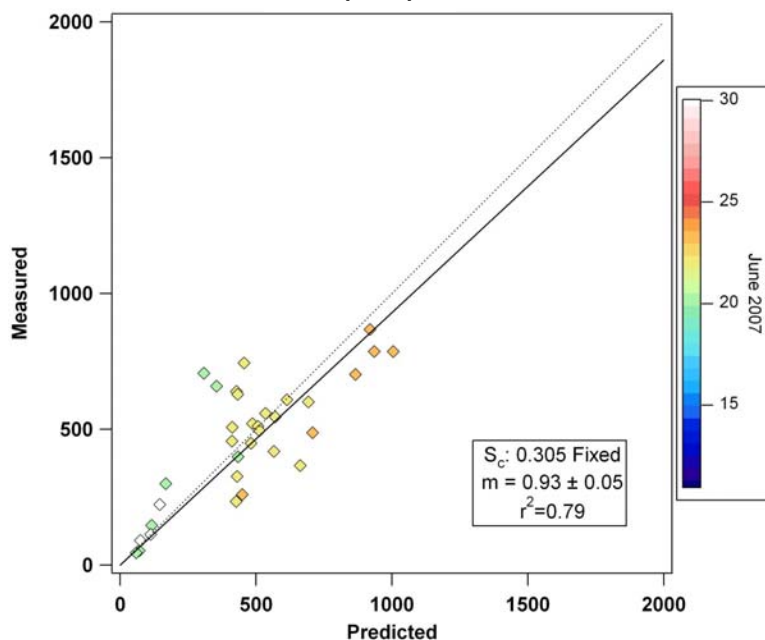
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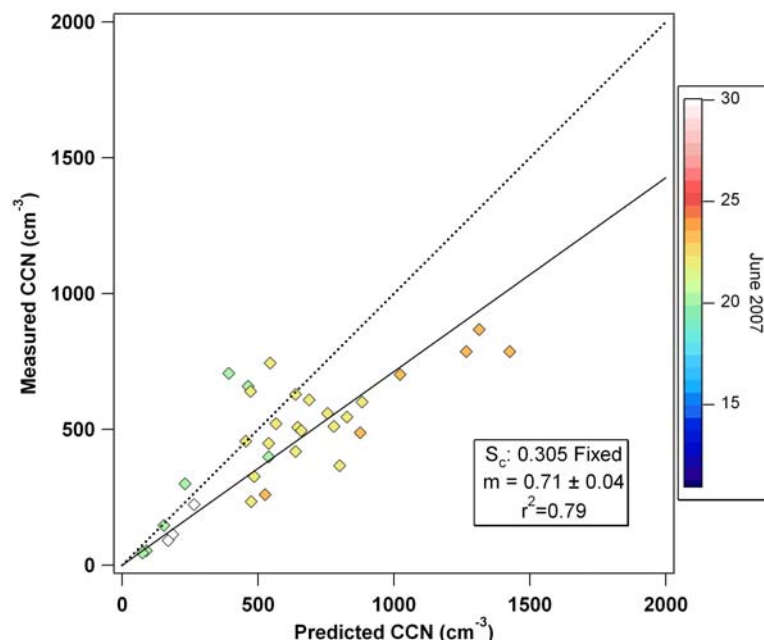
CLASIC 2007

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G(85)

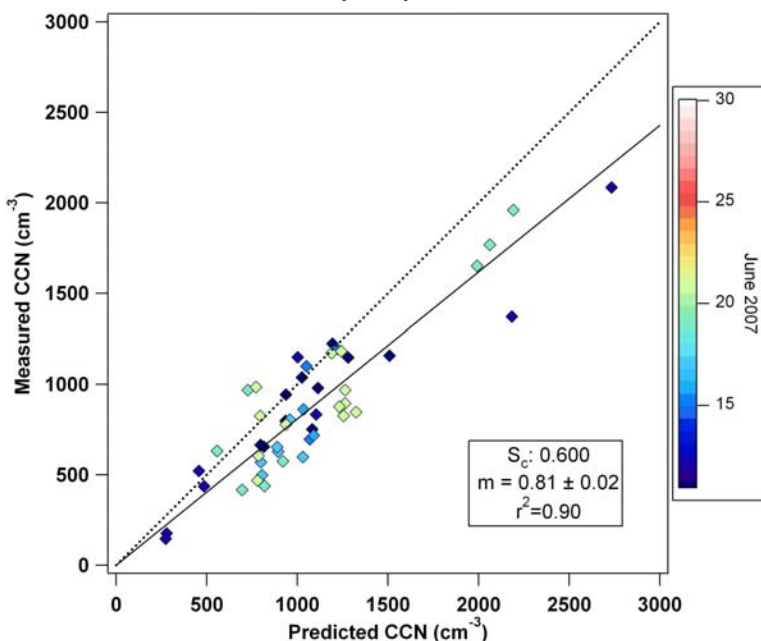


Pure Ammonium Sulfate

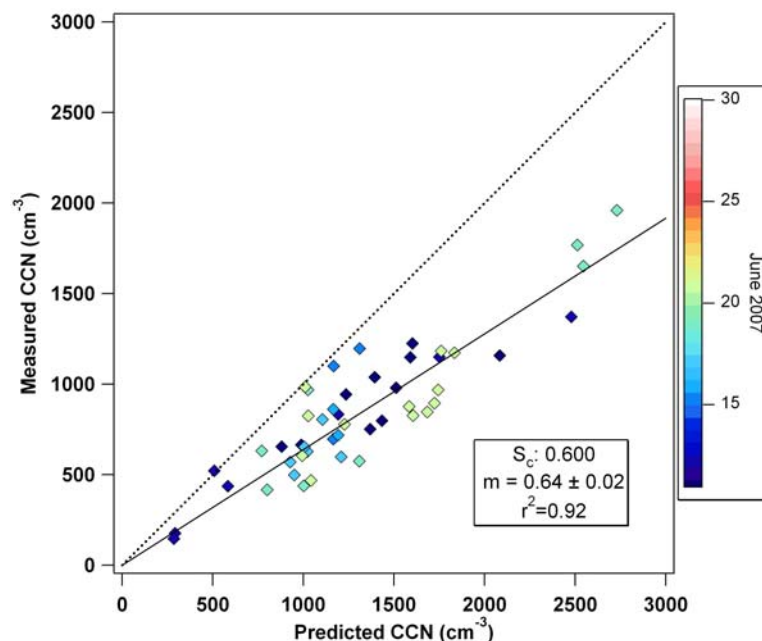


# CCN Scatterplot (0.60)

### G(85)



### Pure Ammonium Sulfate





# Summary

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- During the CLASIC 2007 campaign a DMA/TDMA and DMT CCNc were operated over 14 research flights measuring size distribution, hygroscopic growth distributions, and the number of CCN that would activate at 0.165 %, 0.305%, and 0.600%
- The average size distribution indicate a very stable volume mode and larger concentration within the boundary layer
- The hygroscopic growth distributions indicate the presence of organics
- Predicted CCN concentration agree well with measured CCN concentrations at supersaturations of 0.165% and 0.305%. However over prediction occurs at a supersaturation of 0.600%



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# Thank You

[http://www.met.tamu.edu/research/aerosol/Webpage/CLASIC\\_pages/CLASICcalendar607.html](http://www.met.tamu.edu/research/aerosol/Webpage/CLASIC_pages/CLASICcalendar607.html)



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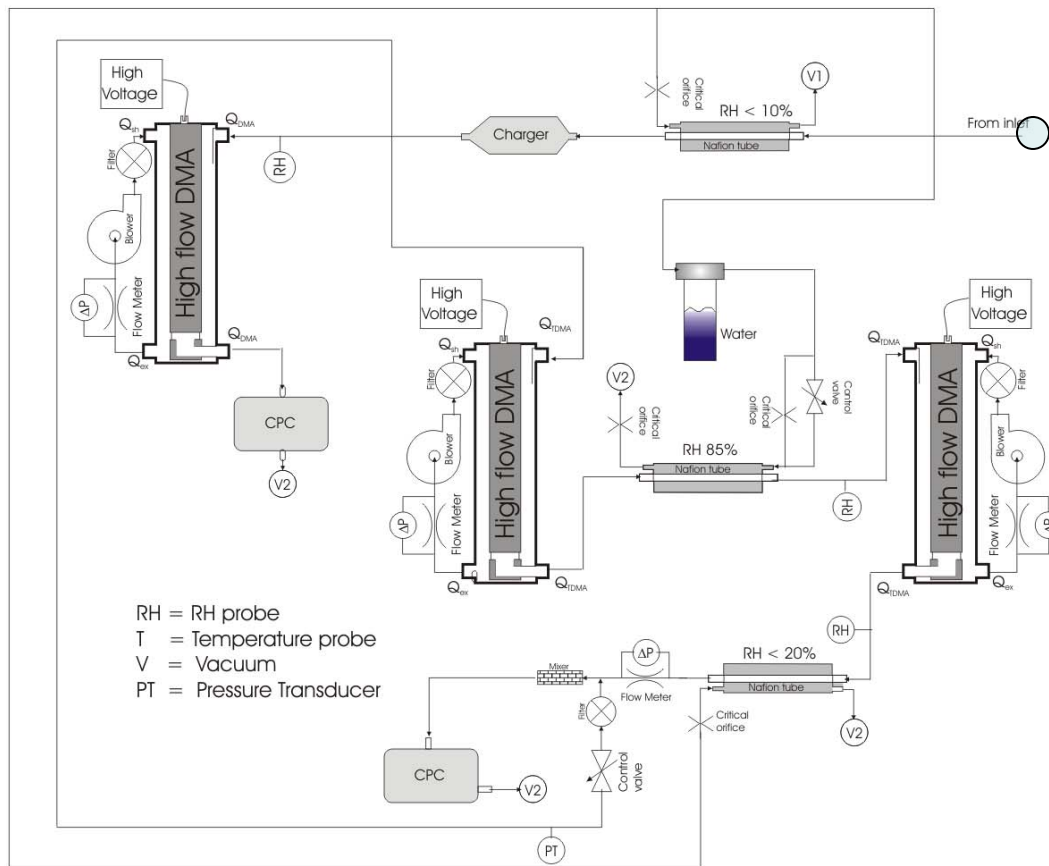


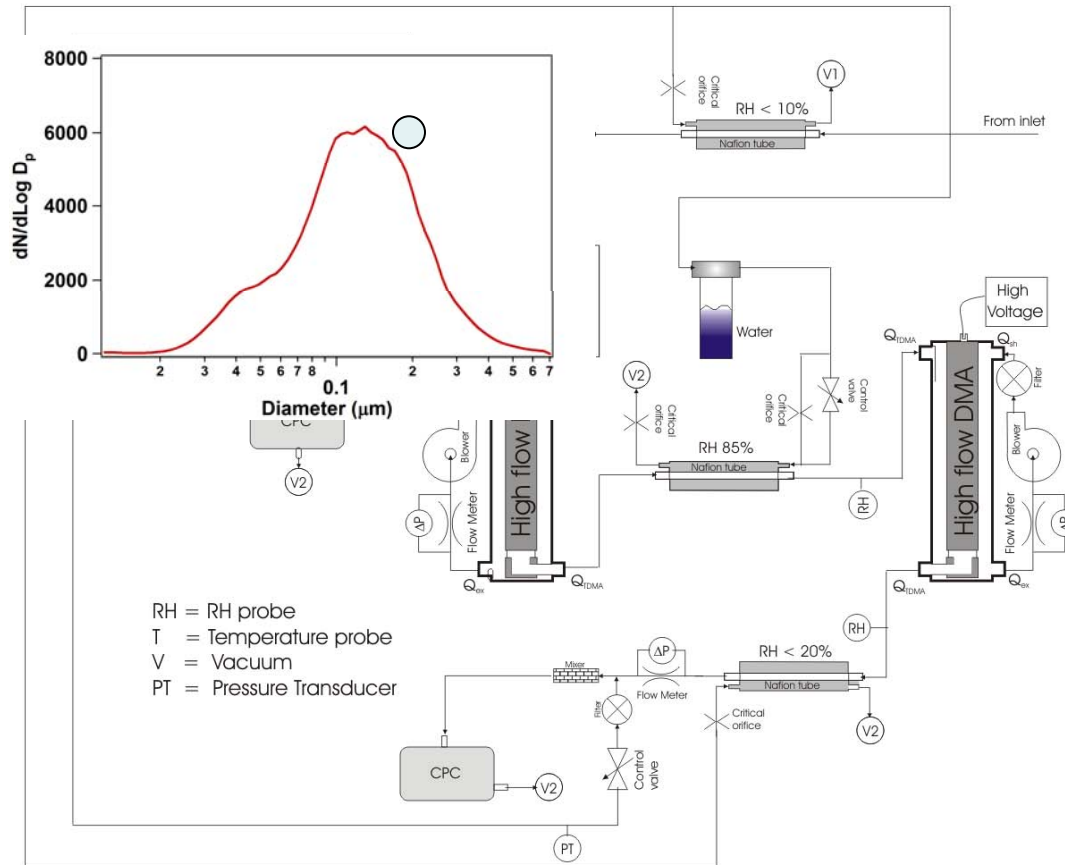
# Extra Slides

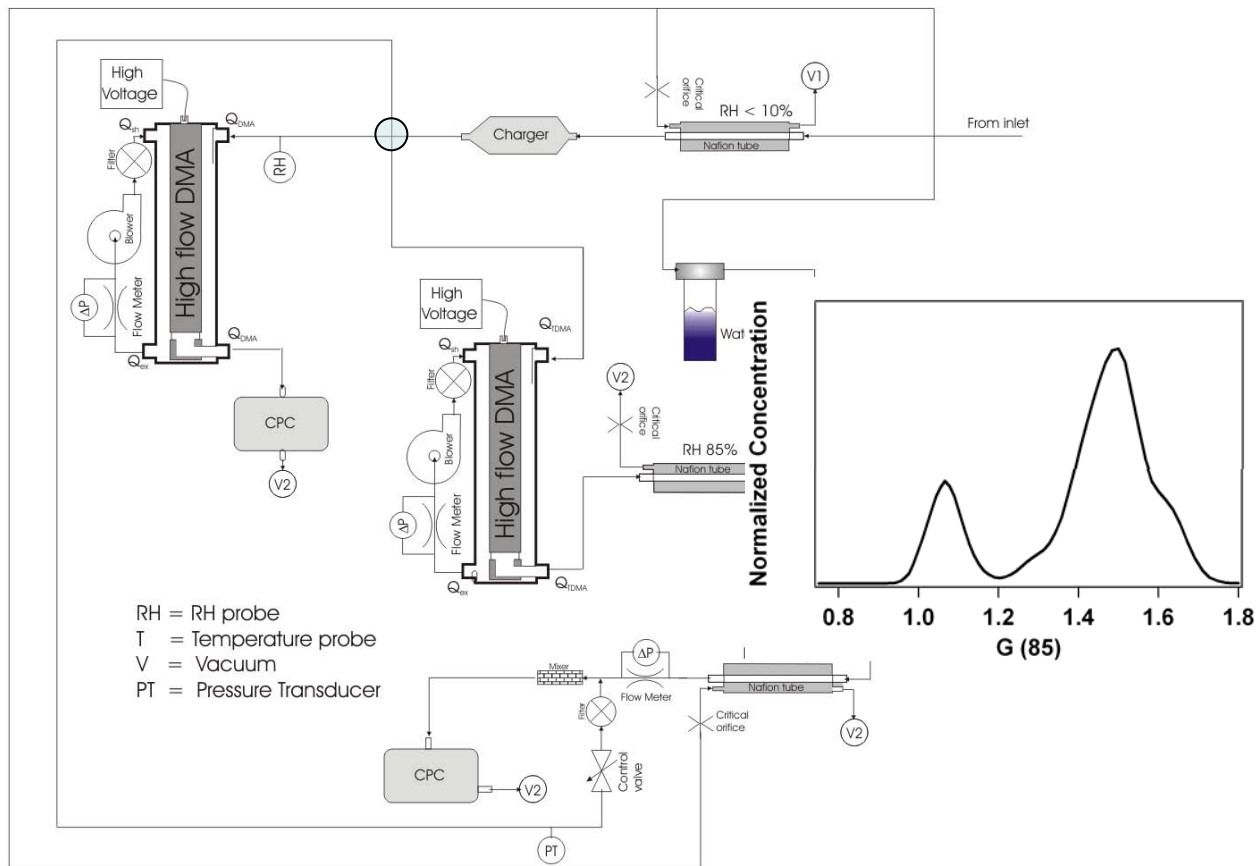


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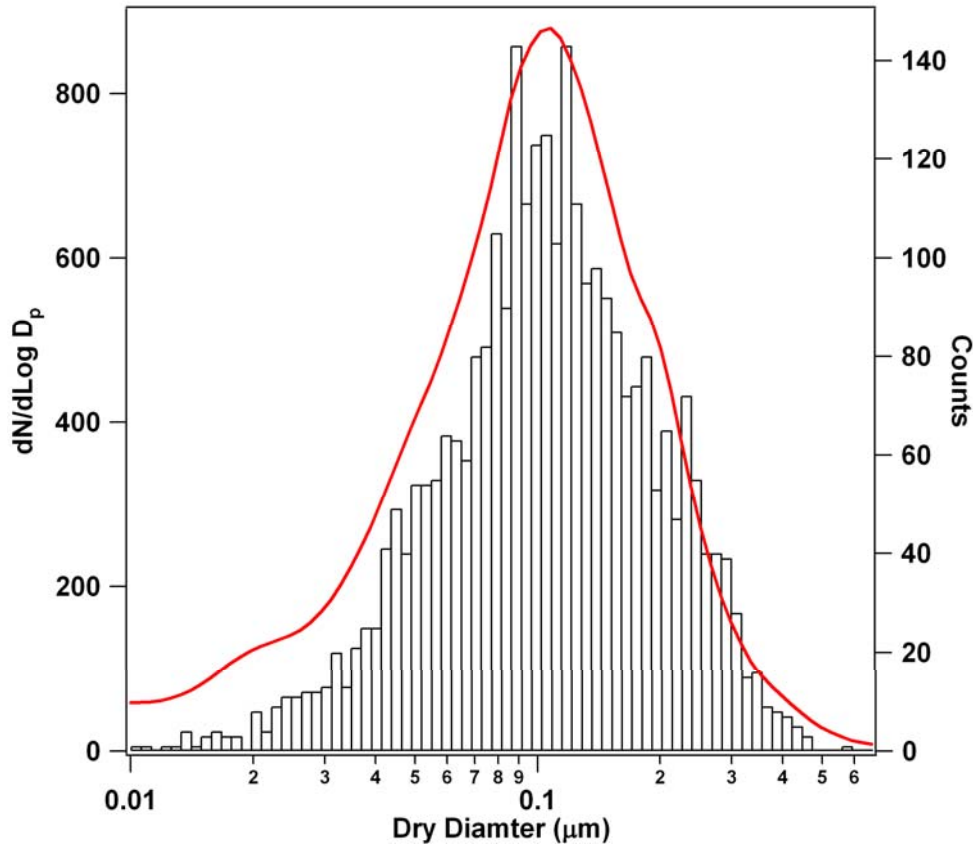








- The size distributions for the DMA are recovered with a Twomey Algorithm



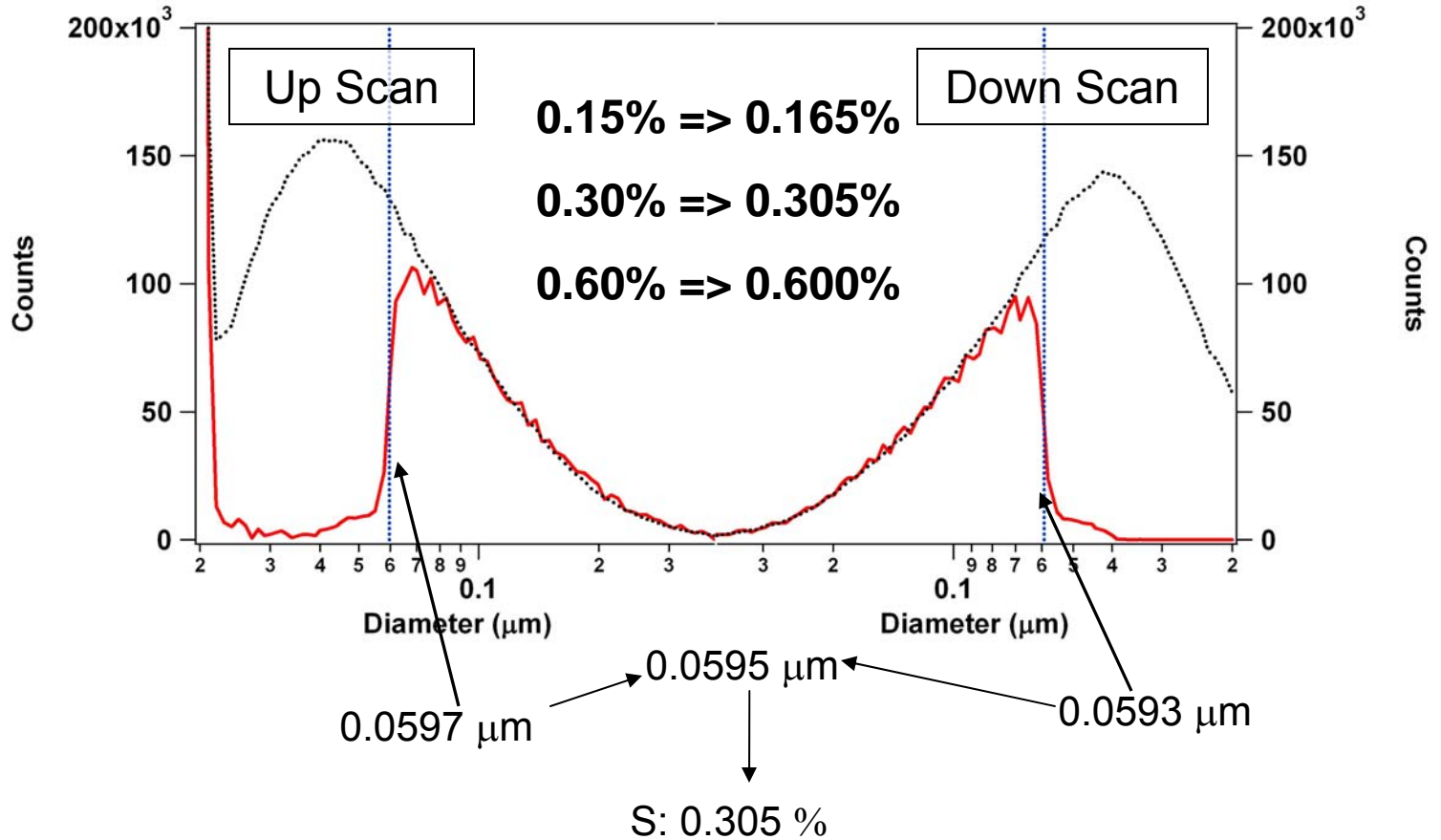
Integrated Concentration  
 Number: 591.4 /cm<sup>3</sup>  
 Surface: 36.8 µm<sup>2</sup>/cm<sup>3</sup>  
 Volume: 1.5 µm<sup>3</sup>/cm<sup>3</sup>



# CCNc Calibration

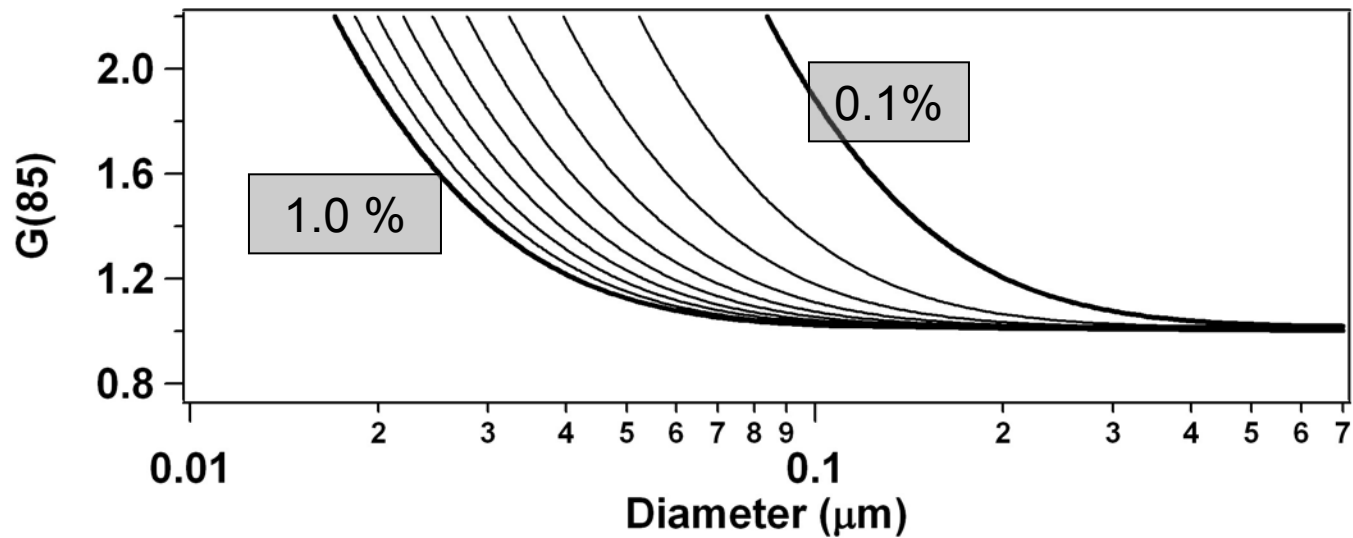
- Atomized ammonium sulfate is passed through a DMA
  - 0.020 up to 0.400  $\mu\text{m}$
  - 15 minute scan time
- 50% point is determined to be the activation diameter
- Kohler theory is used to calculate the corresponding supersaturation

# CCNc Calibration



# CCNc Calibration

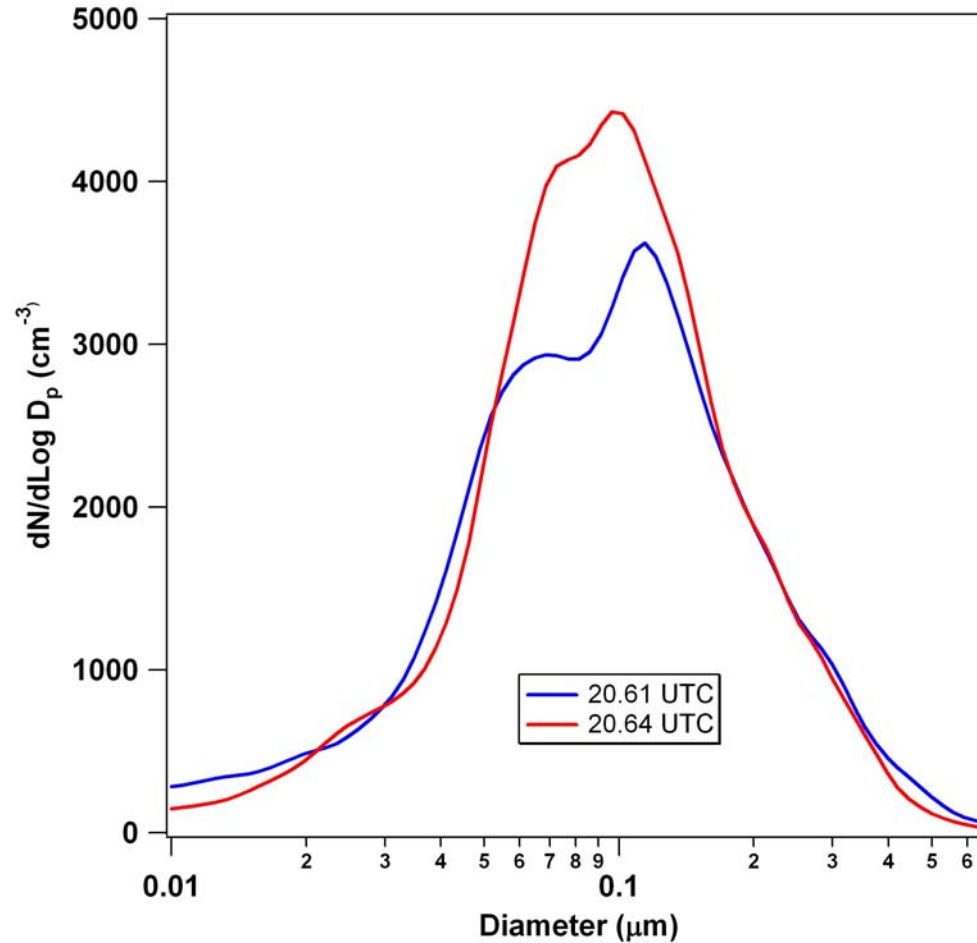
One decadal range of Sc values for Ammonium Sulfate



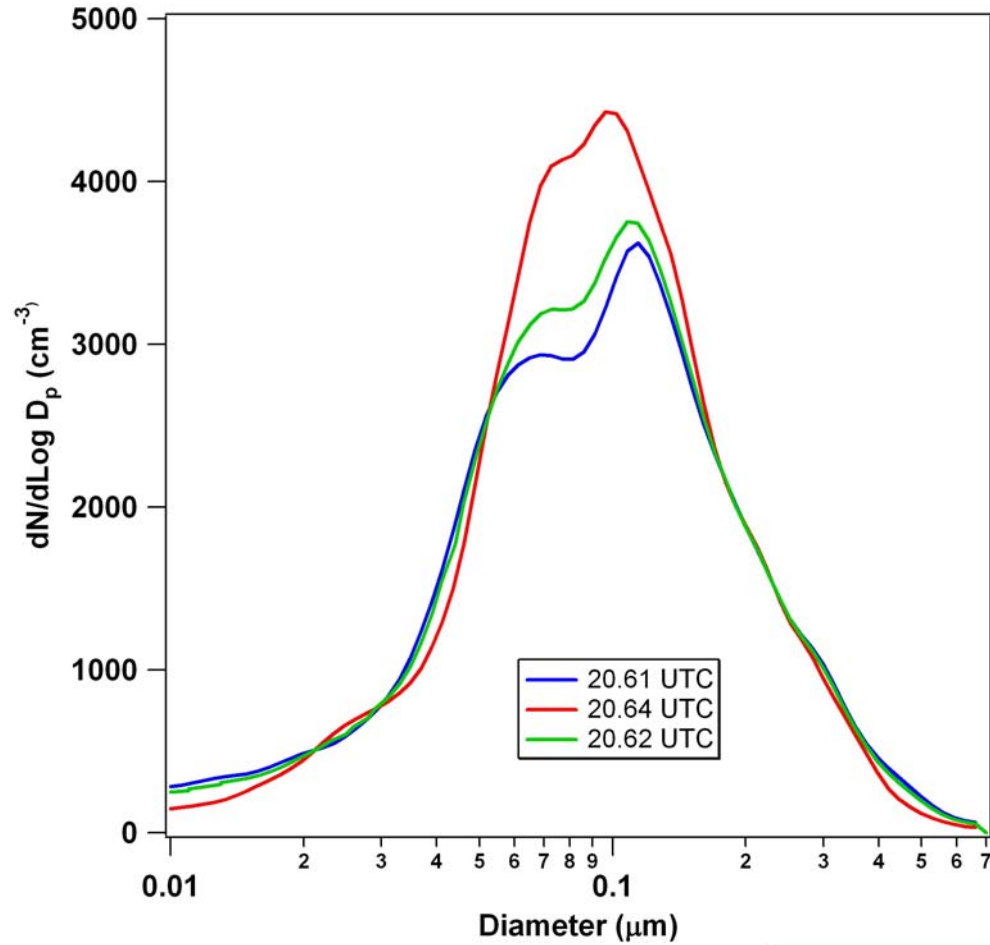
# CCNc Calibration

- Example from June 19<sup>th</sup> at 20:37 UTC
- DMT CCNc measured roughly 1353 +/- 230 cm<sup>-3</sup> activated at a supersaturation of 0.305%

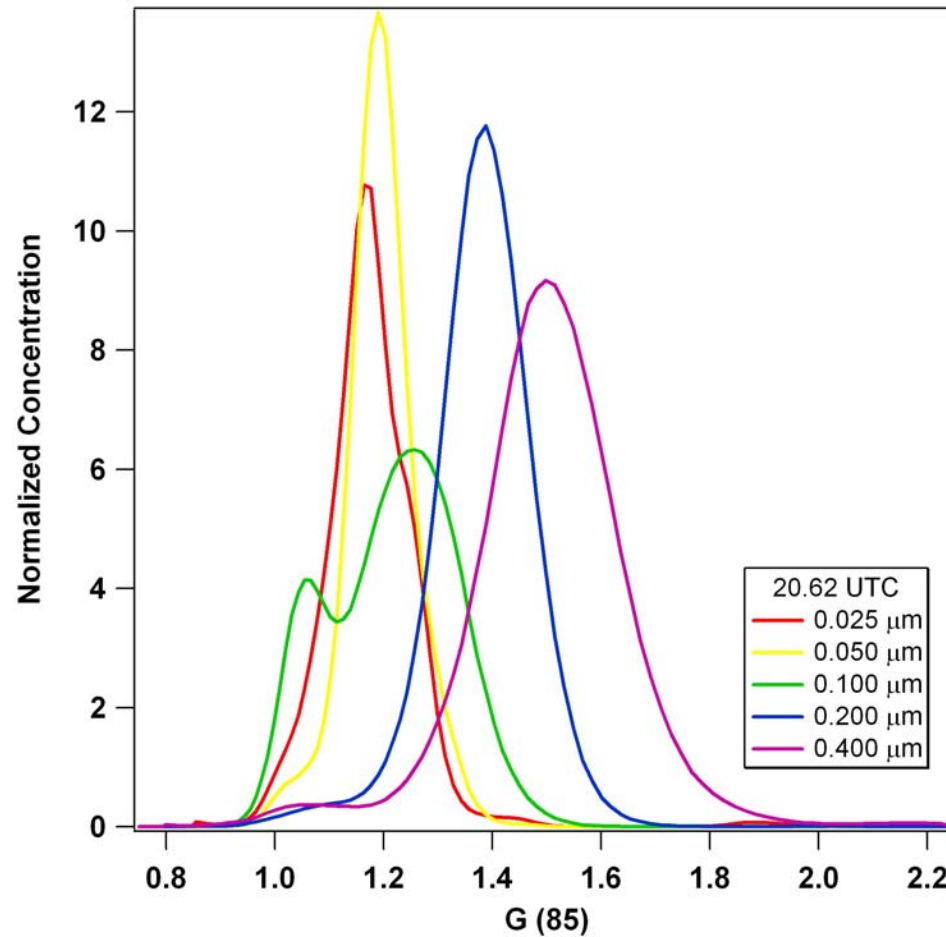
# CCNc Calibration



# CCNc Calibration

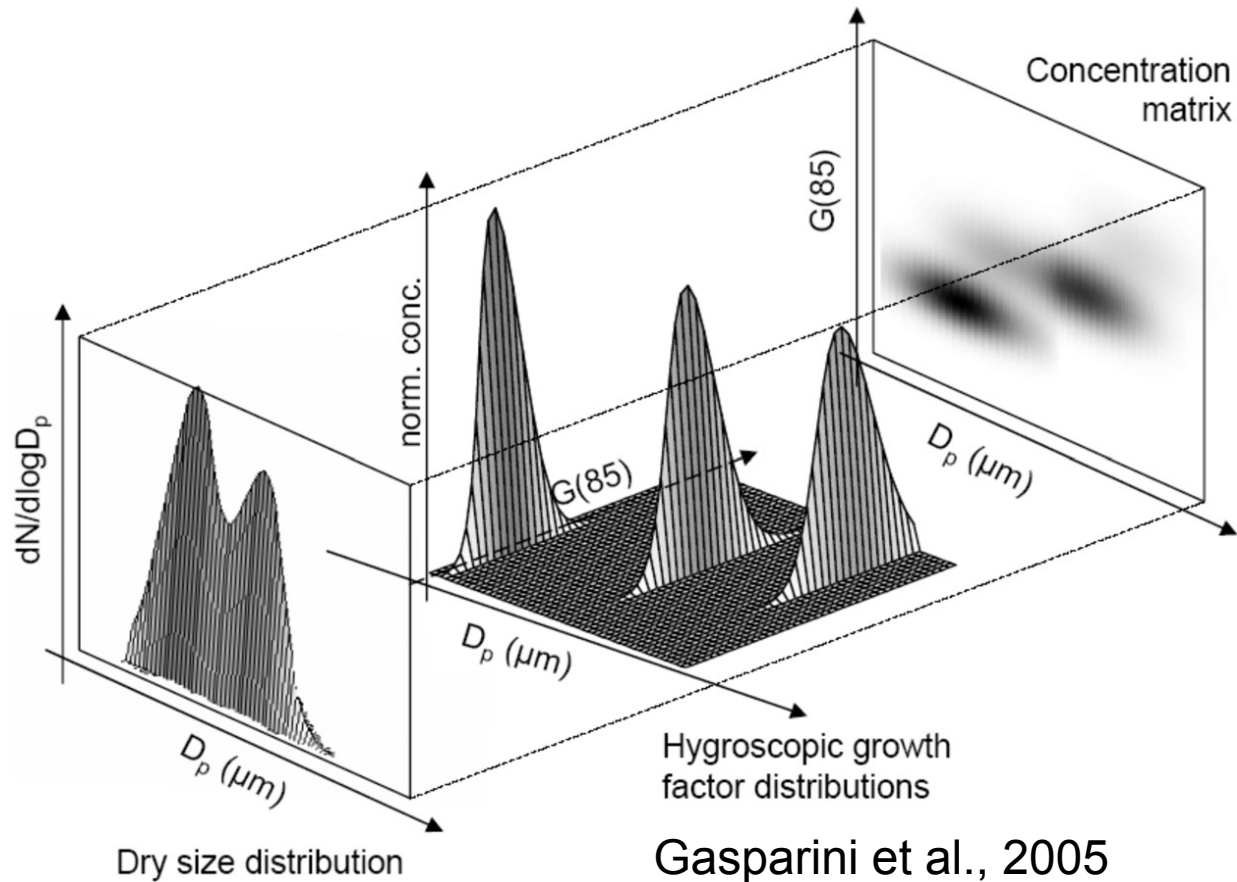


# CCNc Calibration

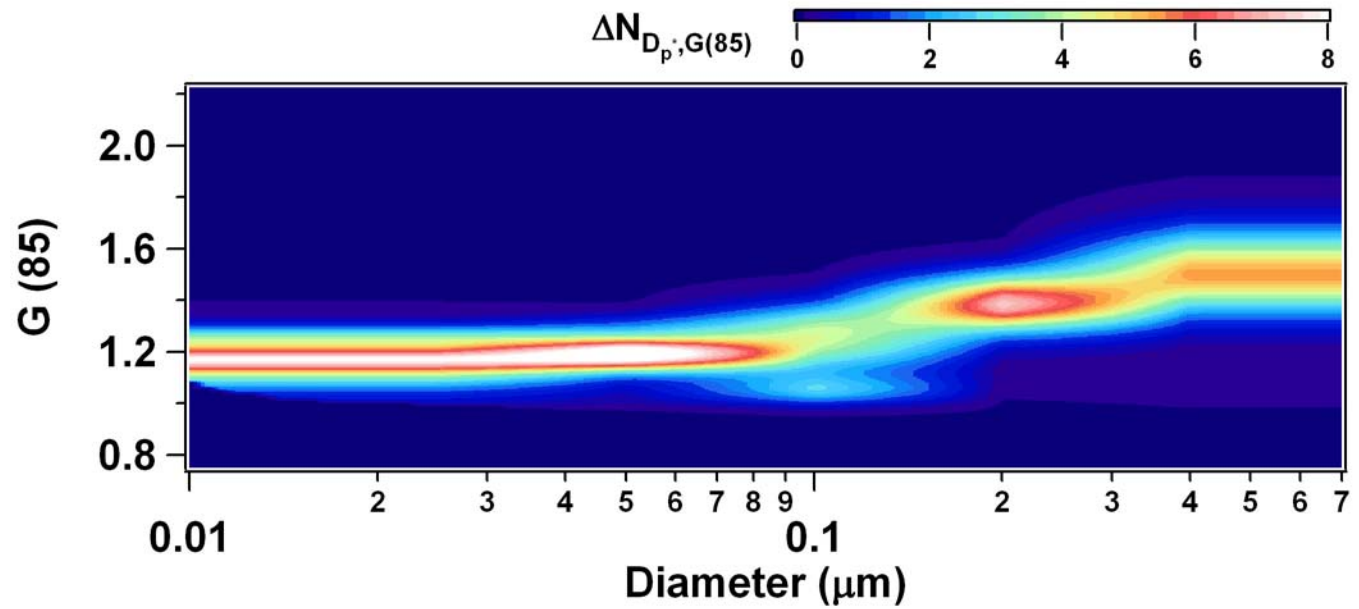




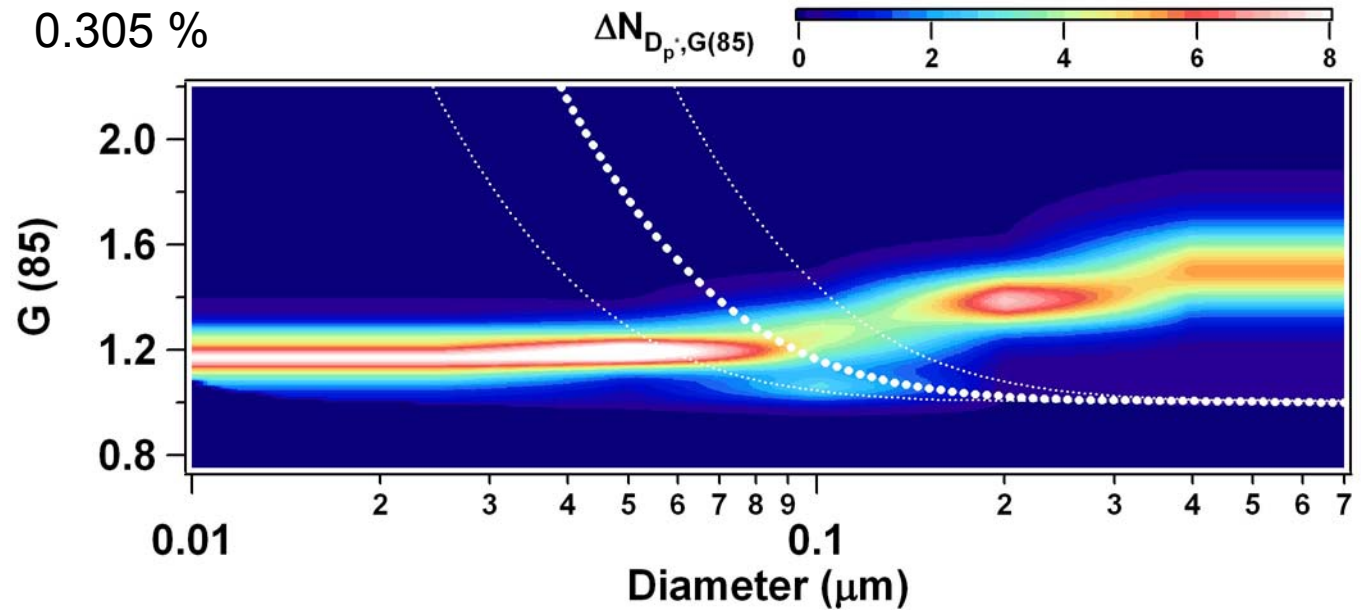
# CCNc Calibration



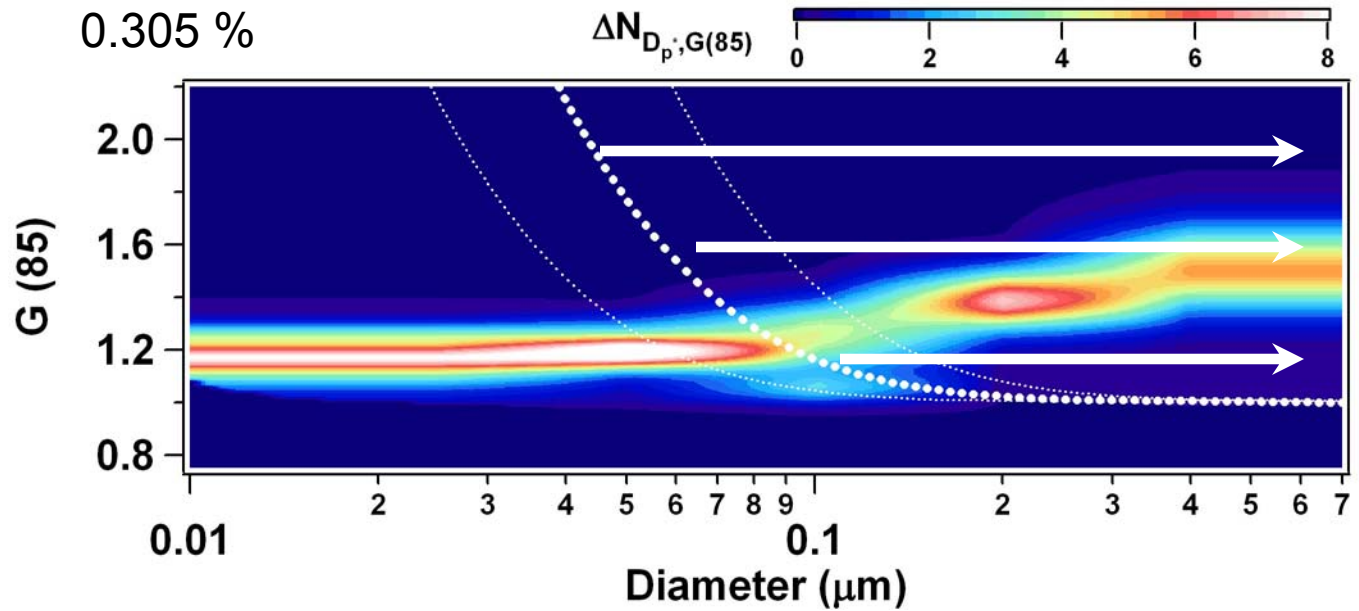
# CCNc Calibration



# CCNc Calibration

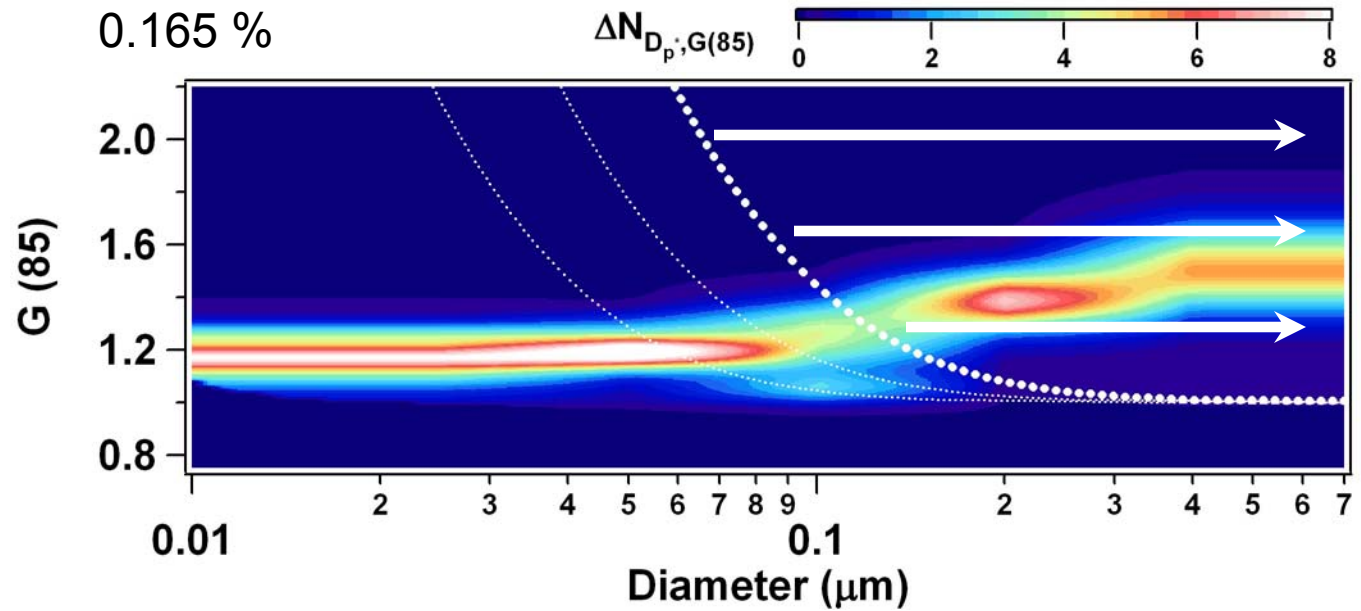


# CCNc Calibration

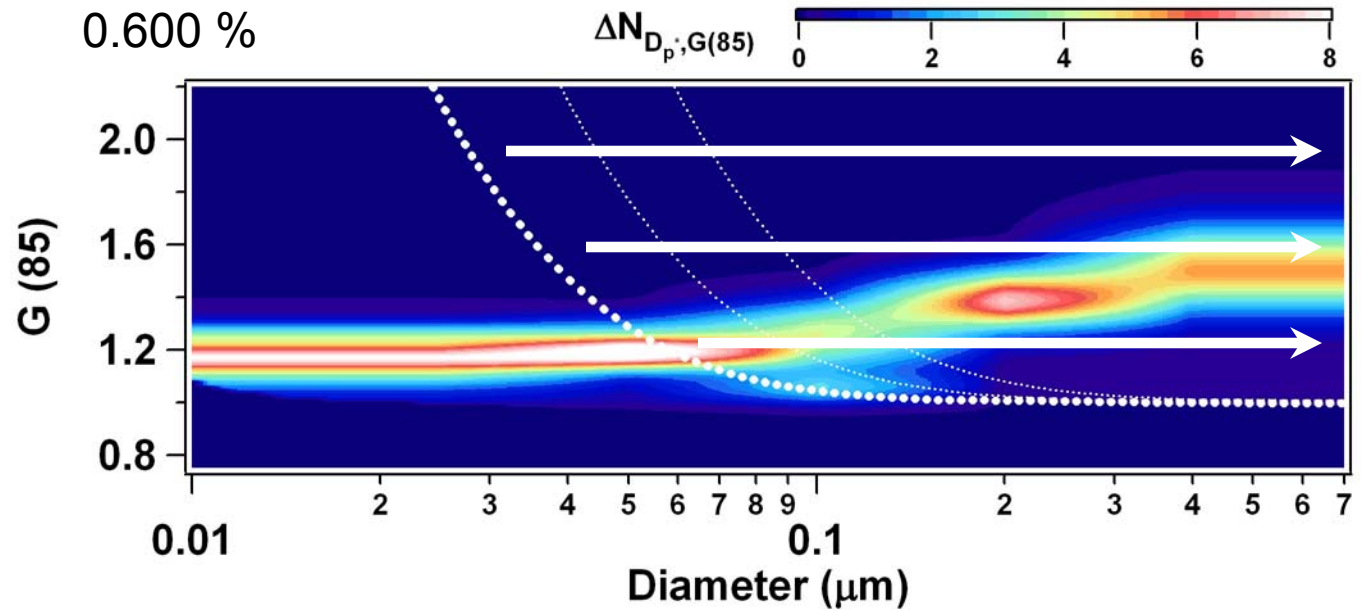


Predicted CCN Concentration:  $1333 \text{ cm}^{-3}$

# CCNc Calibration



# CCNc Calibration



# Growth Factor Prediction

Köhler Equation  $\longrightarrow$   $0.85 = \alpha_w \exp \left[ \frac{4M_w \sigma_{as}}{RT \rho_w (G(85) * D_p^*)} \right]$



# Growth Factor Prediction

Köhler Equation  $\longrightarrow$   $0.85 = \alpha_w \exp \left[ \frac{4M_w \sigma_{as}}{RT \rho_w (G(85)^* D_p^*)} \right]$

$$V_{85} = \frac{\pi}{6} D_{p(85)}^3 = V_i + V_{as}$$

$$V_{dry} = \frac{\pi}{6} D_p^{*3} = V_i + V_s$$

$$\frac{m_s}{\epsilon_s \rho_{as}} - \frac{m_s}{\rho_s} = \frac{\pi}{6} (D_{p(85)}^3 - D_p^{*3})$$

# Growth Factor Prediction

Köhler Equation  $\longrightarrow$   $0.85 = \alpha_w \exp \left[ \frac{4M_w \sigma_{as}}{RT \rho_w (G(85)^* D_p^*)} \right]$

$$V_{85} = \frac{\pi}{6} D_{p(85)}^3 = V_i + V_{as}$$

$$V_{dry} = \frac{\pi}{6} D_p^{*3} = V_i + V_s$$

$$\frac{m_s}{\varepsilon_s \rho_{as}} - \frac{m_s}{\rho_s} = \frac{\pi}{6} (D_{p(85)}^3 - D_p^{*3})$$

Surface Tension  $\longrightarrow$   $\sigma_{as} = \sigma_w(T) + \frac{2 \times 10^{-6} \varepsilon_s \rho_{as}}{M_s}$

# Growth Factor Prediction

Köhler Equation  $\longrightarrow$   $0.85 = \alpha_w \exp \left[ \frac{4M_w \sigma_{as}}{RT \rho_w (G(85)^* D_p^*)} \right]$

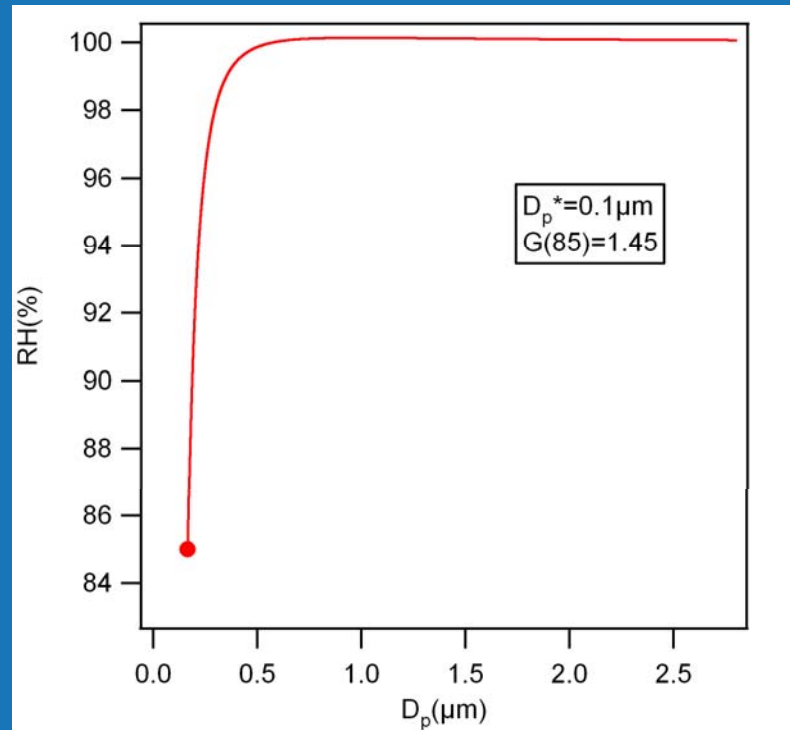
$$V_{85} = \frac{\pi}{6} D_{p(85)}^3 = V_i + V_{as}$$

$$V_{dry} = \frac{\pi}{6} D_p^{*3} = V_i + V_s$$

$$\frac{m_s}{\epsilon_s \rho_{as}} - \frac{m_s}{\rho_s} = \frac{\pi}{6} (D_{p(85)}^3 - D_p^{*3})$$

Surface Tension  $\longrightarrow$   $\sigma_{as} = \sigma_w(T) + \frac{2 \times 10^{-6} \epsilon_s \rho_{as}}{M_s}$

# Growth Factor Prediction



TDMA  
Köhler Equation



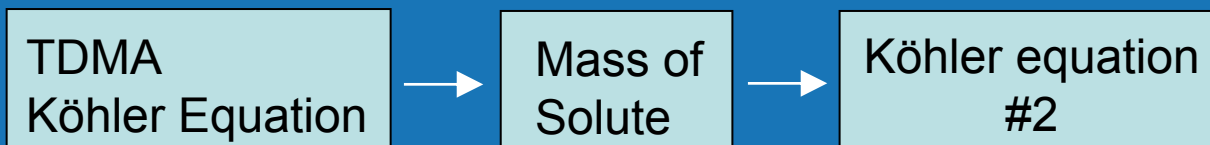
Mass of  
Solute

# Growth Factor Prediction

Köhler Equation #2  $\longrightarrow$   $\frac{e'}{e_s} = \gamma_w \chi_w \exp\left[\frac{4 M_w \sigma_{as}}{RT \rho_w D_d}\right]$

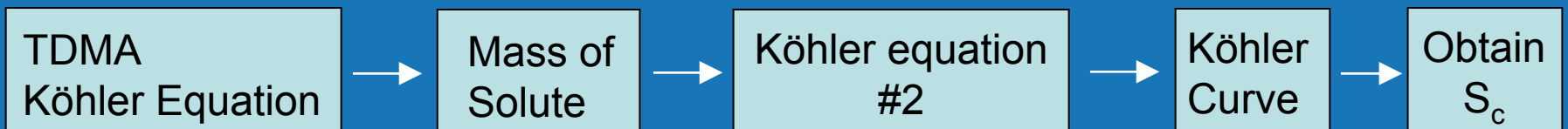
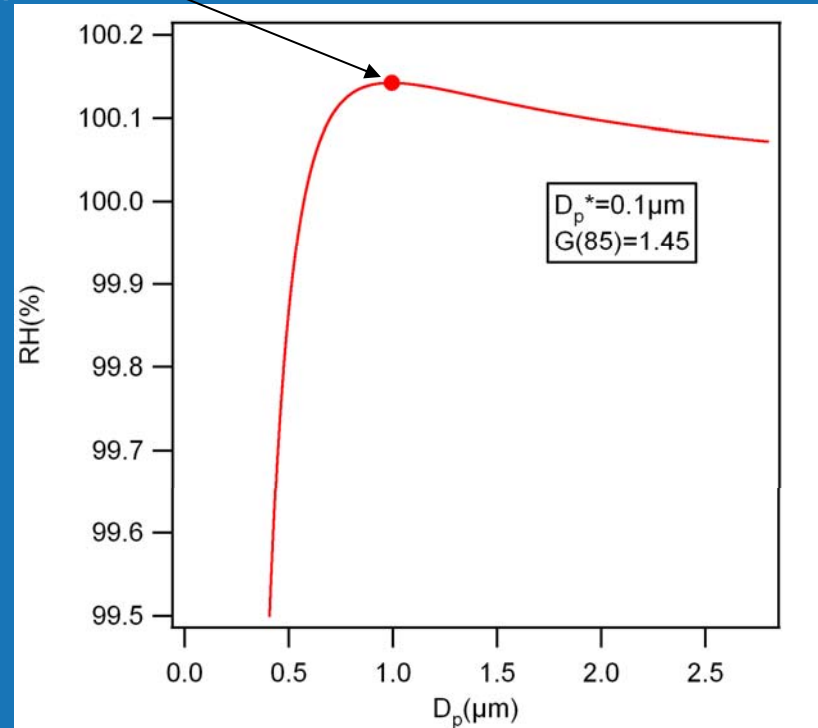
- The soluble mass determined in the initial calculations is fixed in the mass fraction of solute in solution term when calculating  $e'/e_s$ .
- Since this second calculation is representative of a particle in supersaturated conditions, we assume ideal behavior:

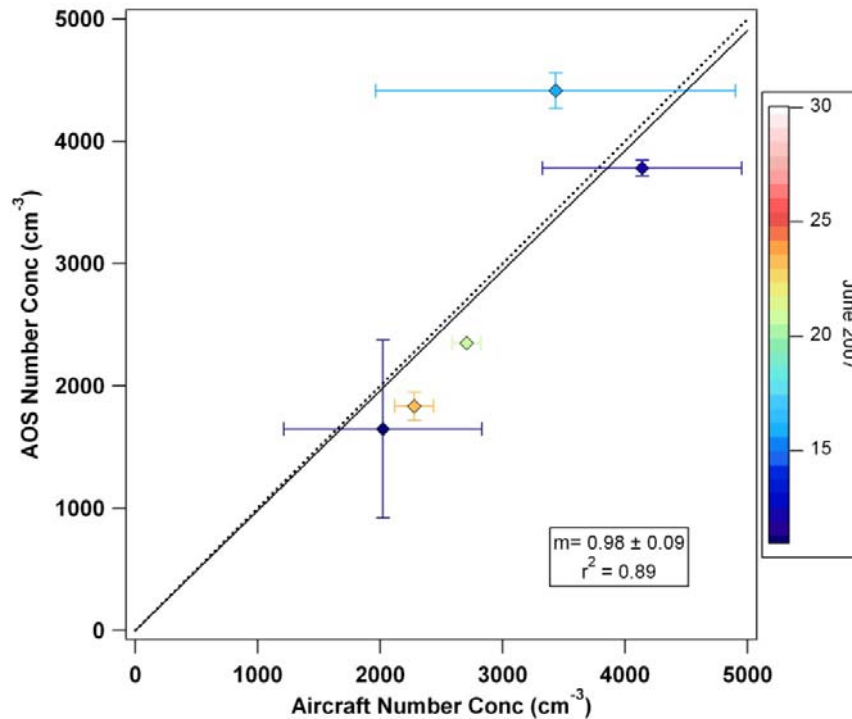
$$\gamma_w = 1 \quad \chi_w = \frac{n_w}{n_w + i n_s} = \frac{1}{\frac{i M_w}{M_s} \left( \frac{\epsilon_s}{1 - \epsilon_s} \right) + 1}$$



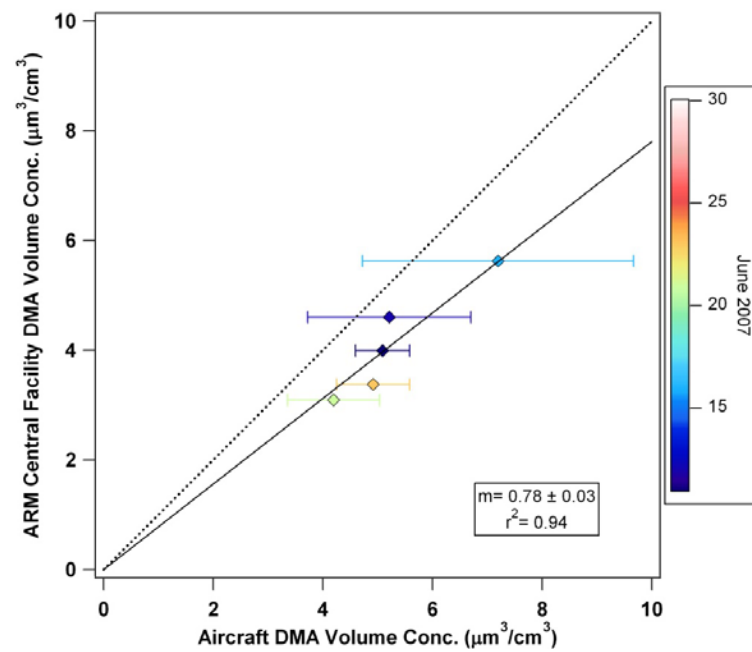
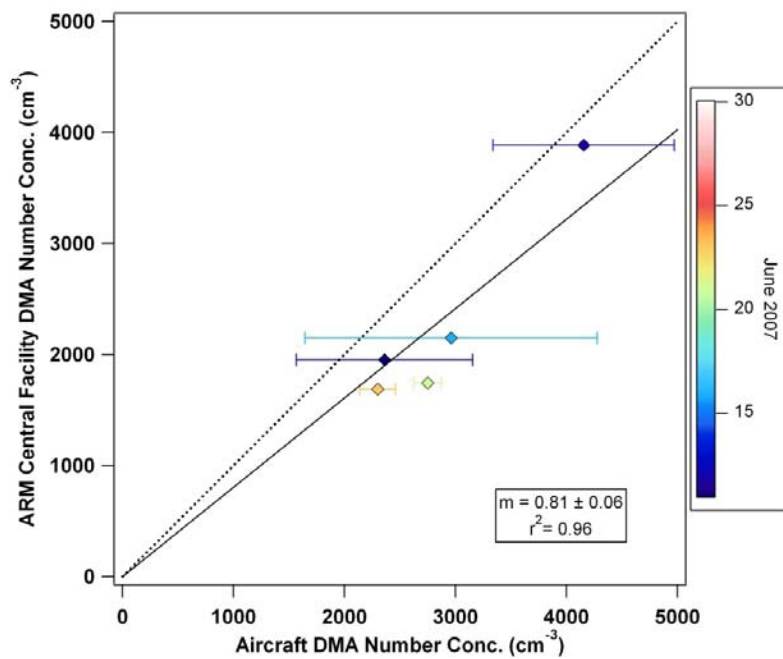
# Growth Factor Prediction

$$\frac{d\left(\frac{e'}{e_s}\right)}{dD_p} = 0$$





- Comparison between integrated number concentration and measured concentration from the AOS TSI 3010





## June 12<sup>th</sup> Morning Flight

